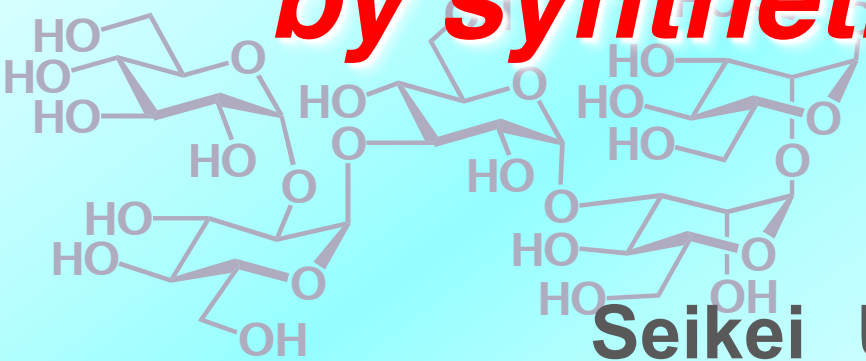


***Glycan recognition revealed
by synthetic chemistry***



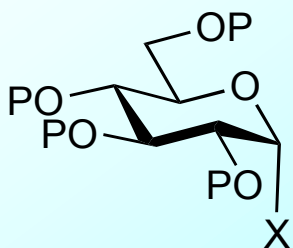
Seikei University
Kiichiro TOTANI

- **Basics of glycosylation reaction**
- **Examples of oligosaccharide synthesis**
- **Development of stereoselective glycosylation reaction**
- **Analysis of glycoprotein quality control system
using chemically synthesized oligosaccharides**

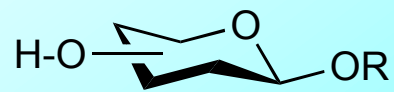
O-Glycosylation reaction

Glycosyl donor

Glycosyl acceptor

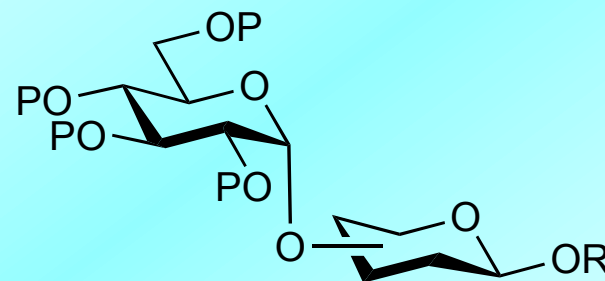


+



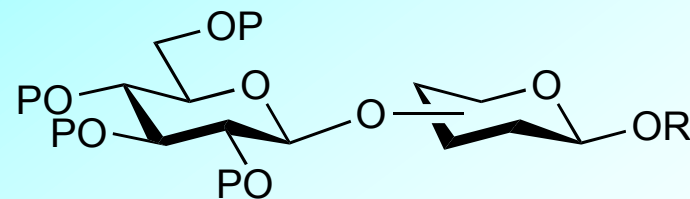
Promoter

Solvent



α -Glycoside

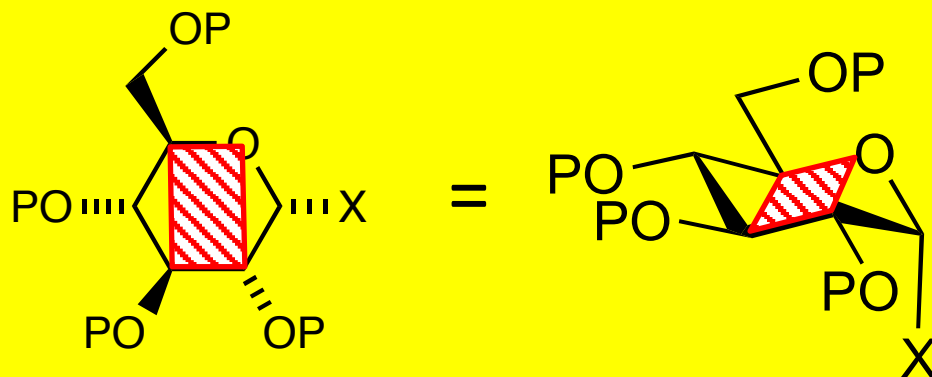
+



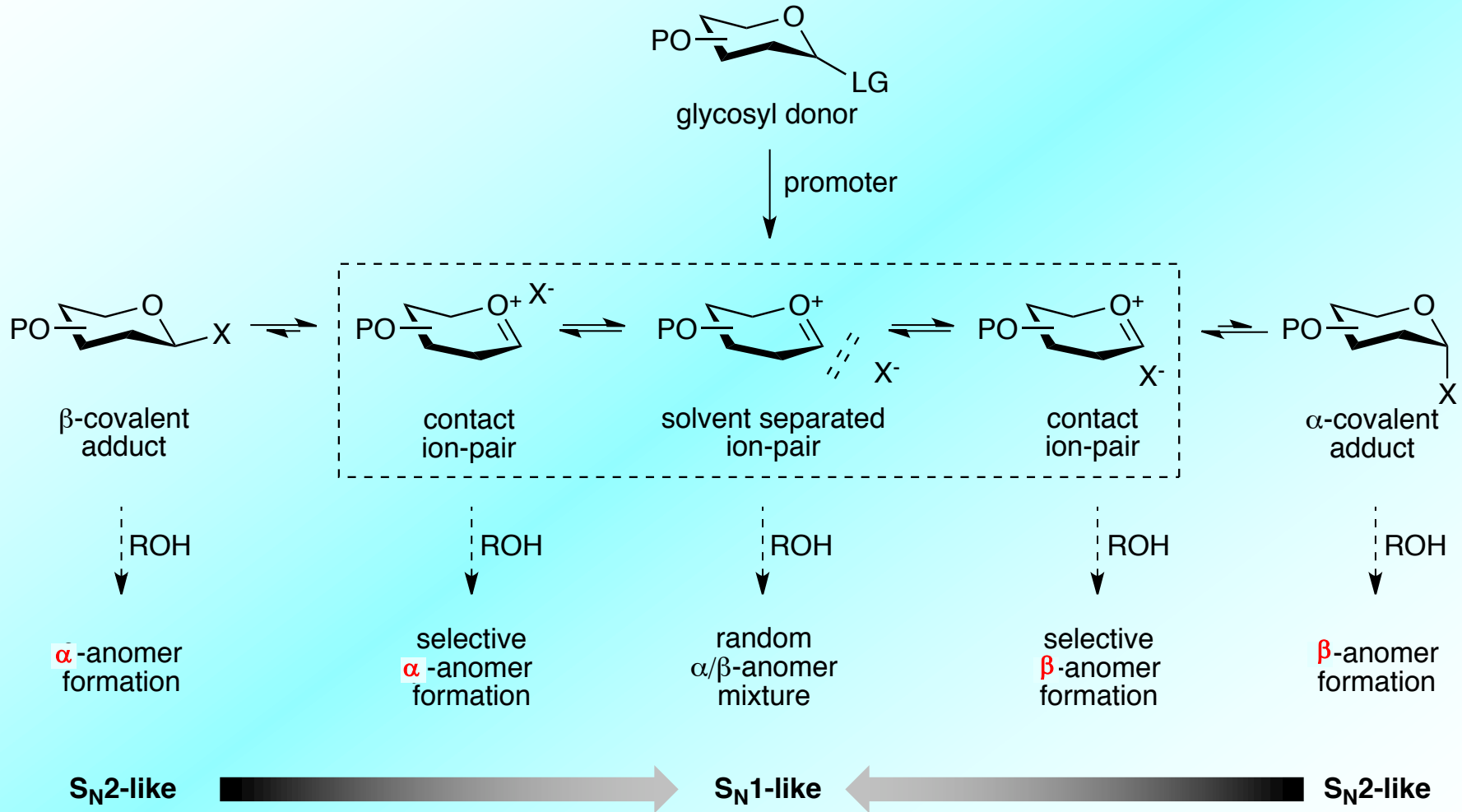
β -Glycoside

P: Protecting group

X: Leaving group

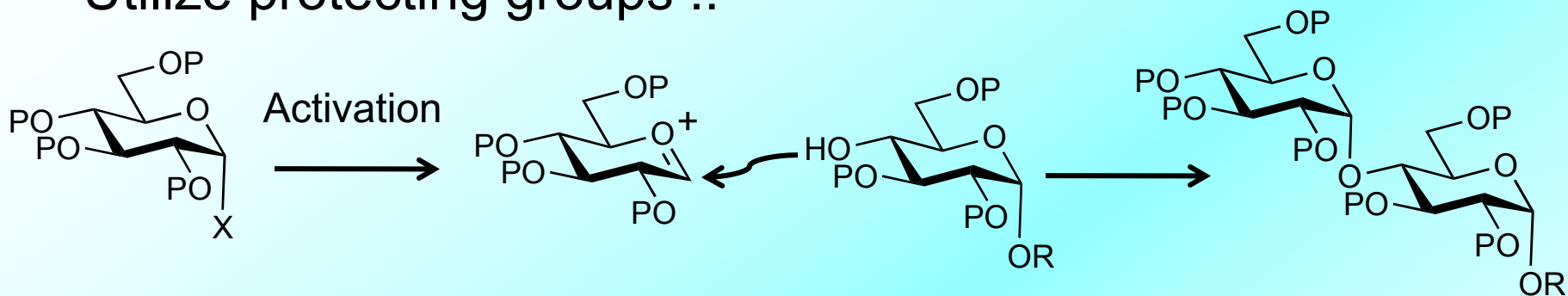


Reaction mechanism of O-glycosylation reaction



To control the regioselectivity of glycosylation

Utilize protecting groups !!

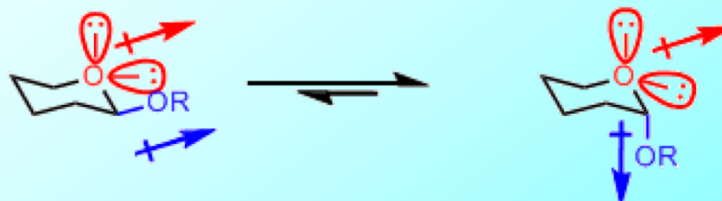


Acyl-Type	Ether-Type	Silyl Ether	Cyclic Acetal
 Ac		 TBDMS	 isopropylidene
	 All	 TBDPS	

To control the stereoselectivity of glycosylation

Glycosyl donor	Preferential stereochemistry		Stereocontrol
	Anomeric effect	Neighboring group participation	
D-Glc D-Gal D-GlcNAc D-GalNAc	α	β	Possible
D-Man	α	α	Difficult
L-Fuc	α	β	Possible
NeuAc	β	none	Difficult

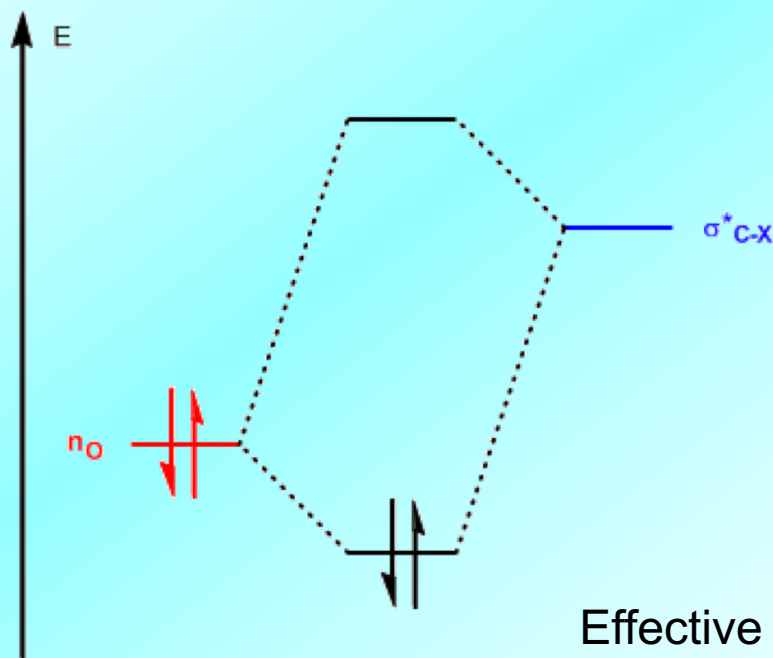
Dipole Minimization



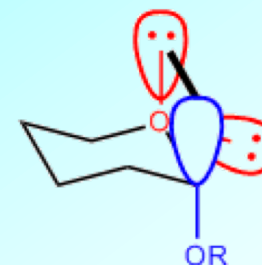
React slowly at a relatively high temperature (RT to c.a. 50° C)

Non-coordinating solvent

Electron Delocalization

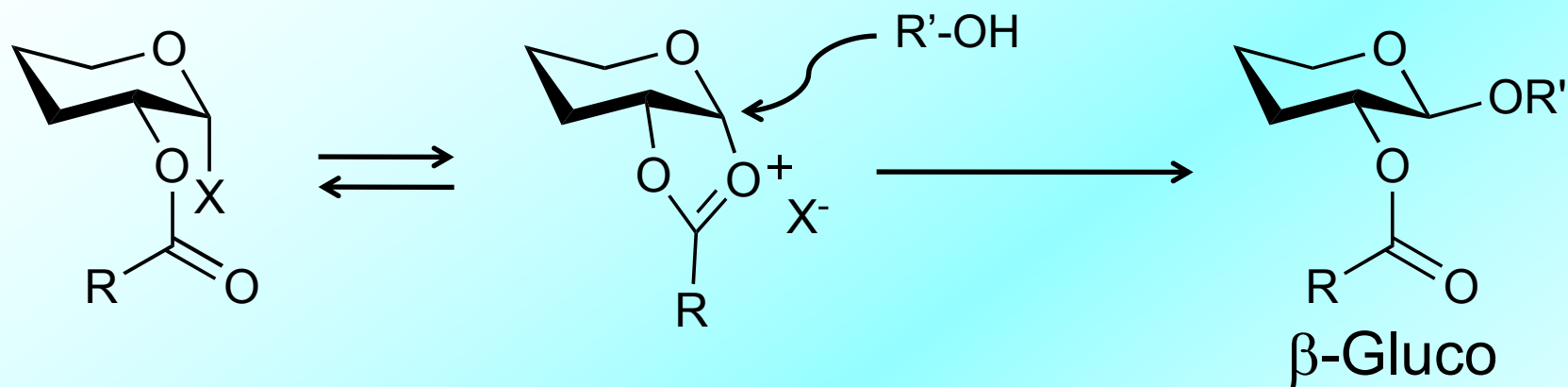


Non-acyl type protecting group at C-2 position

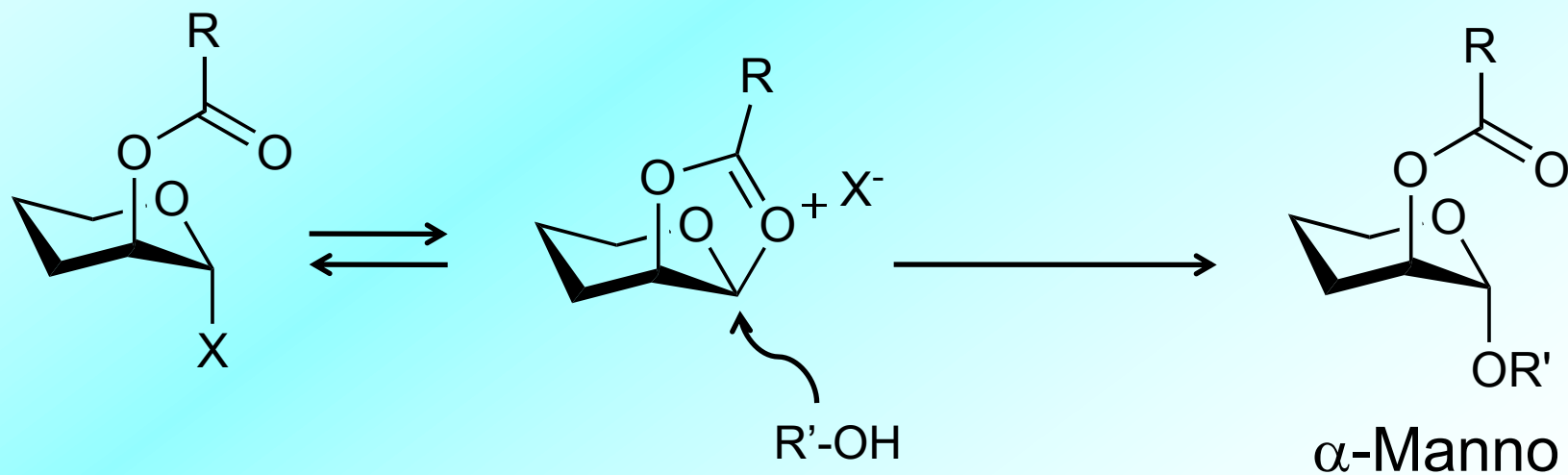


Effective for S_N1-like reaction mechanism

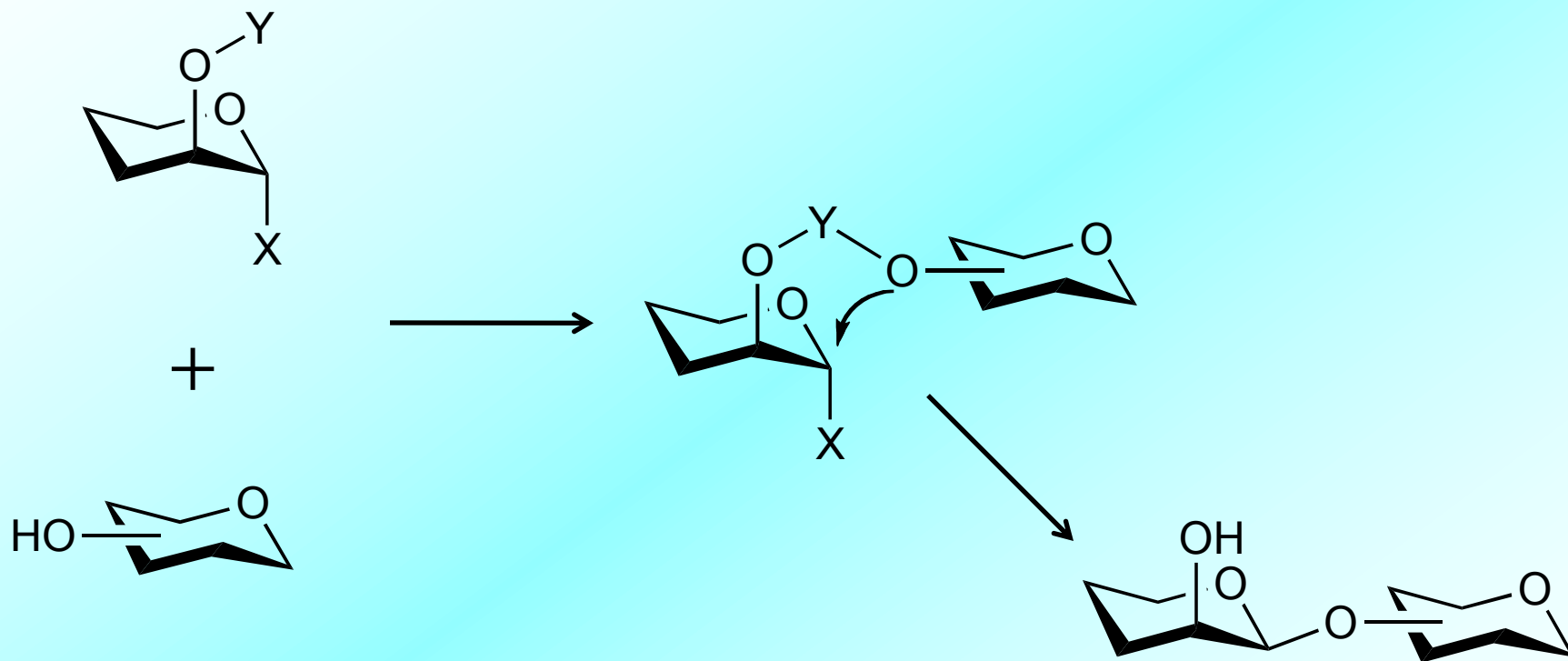
Neighboring group participation



Promoter

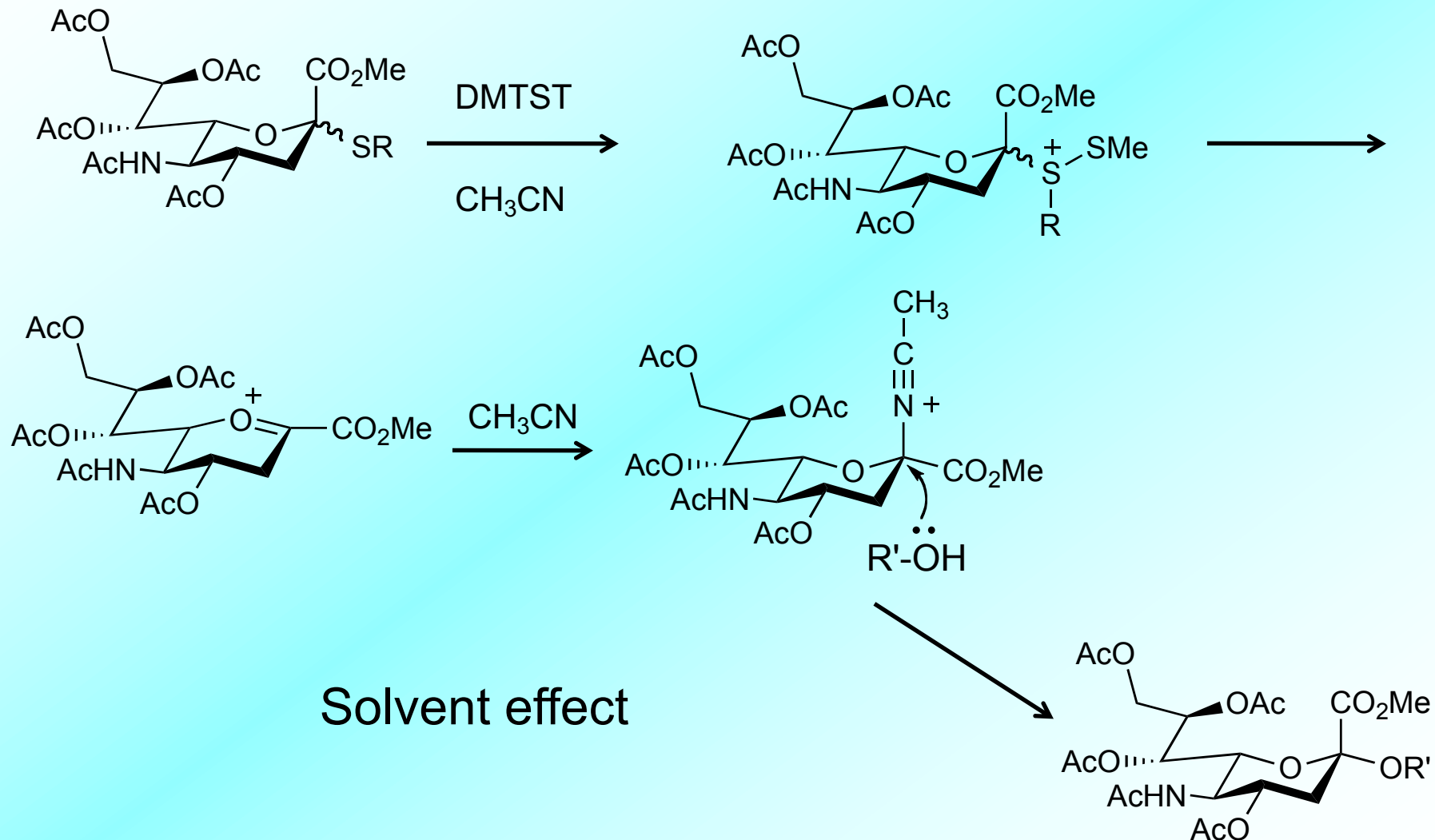


To prepare difficult-to-form β -mannoside

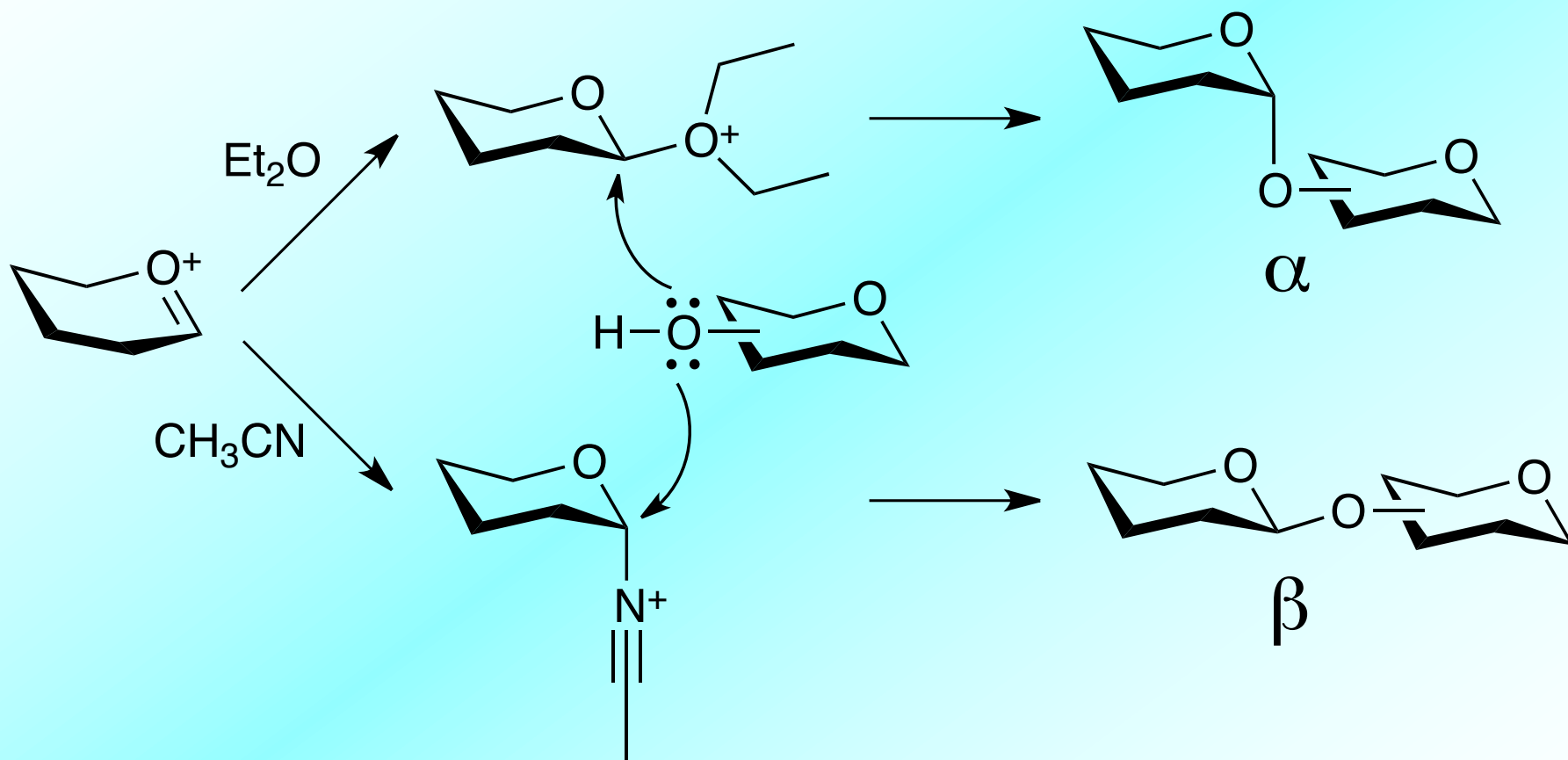


Intramolecular aglycone transfer

To prepare difficult-to-form α -sialoside

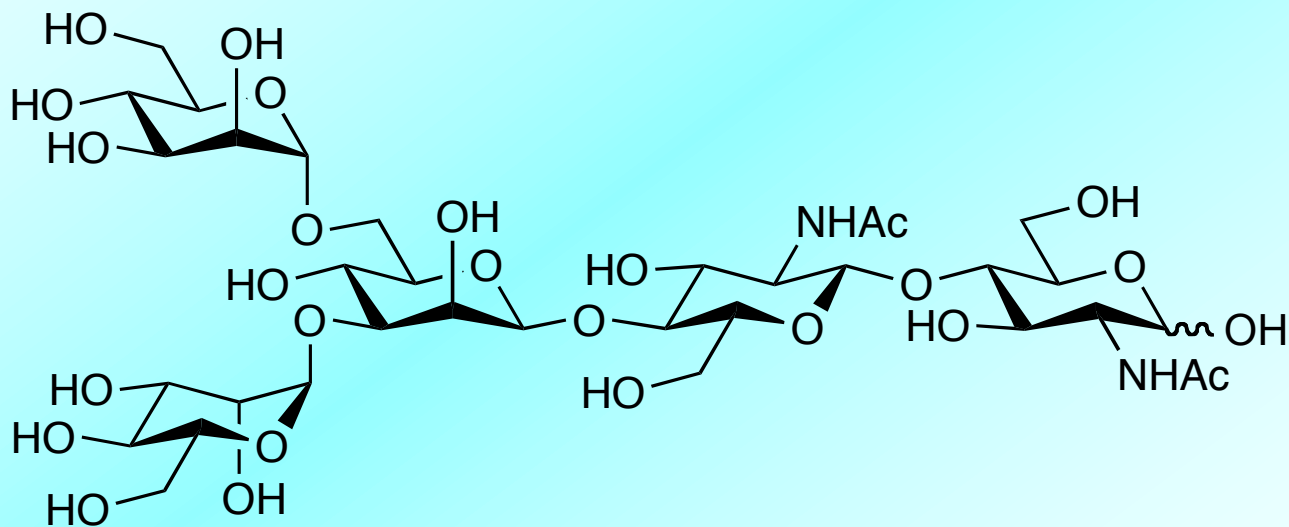


Stereoselective glycosylation by solvent effect



Synthesis of high-mannose-type core pentasaccharide

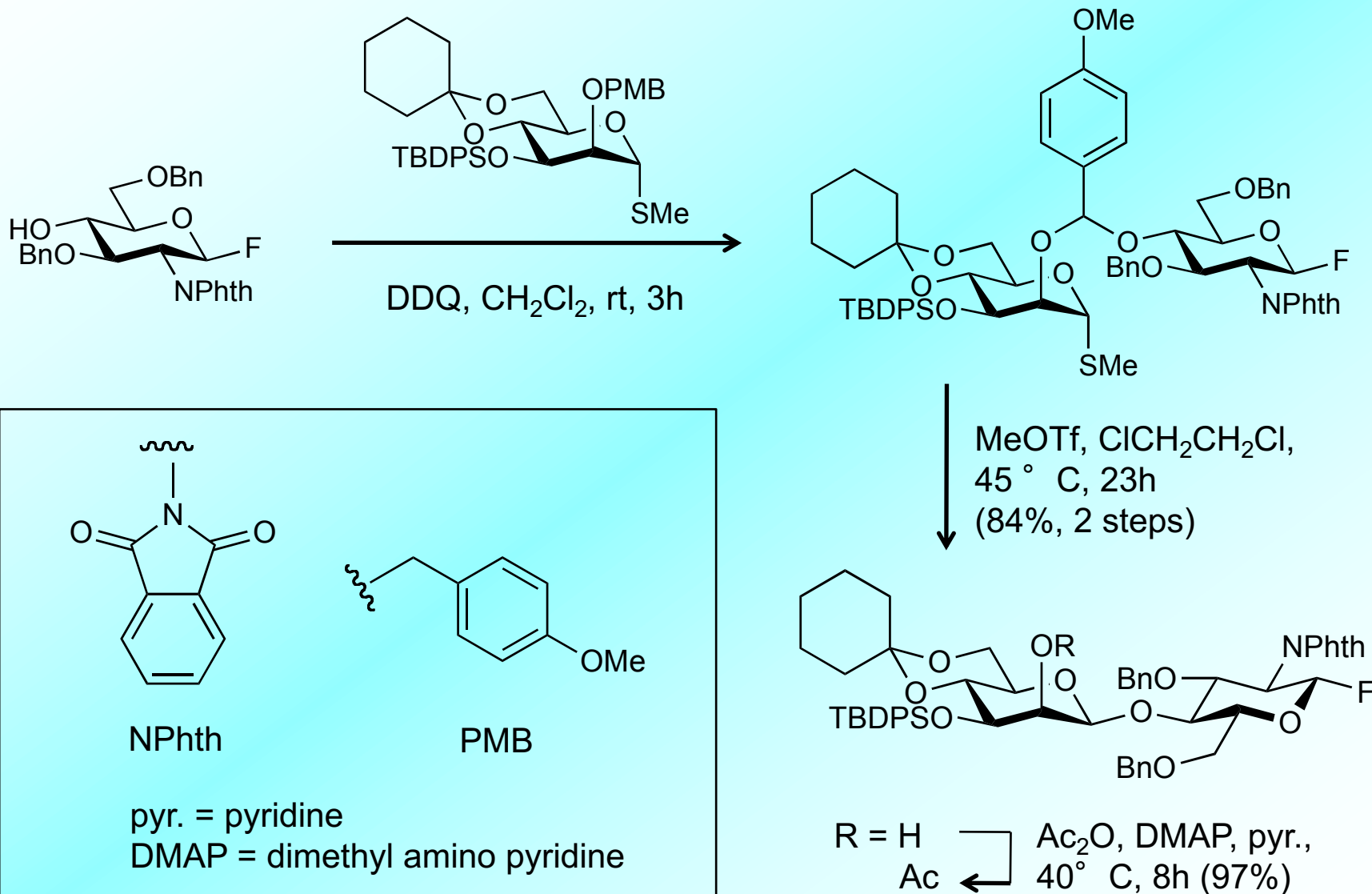
Totani, K.; Matsuo, I.; Ito, Y. *Bioorg. Med. Chem. Lett.* **2004**, 14, 2285.



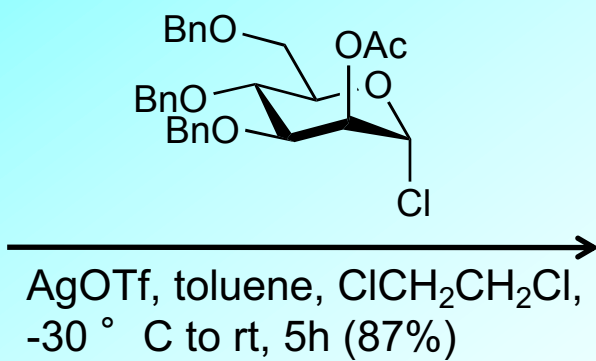
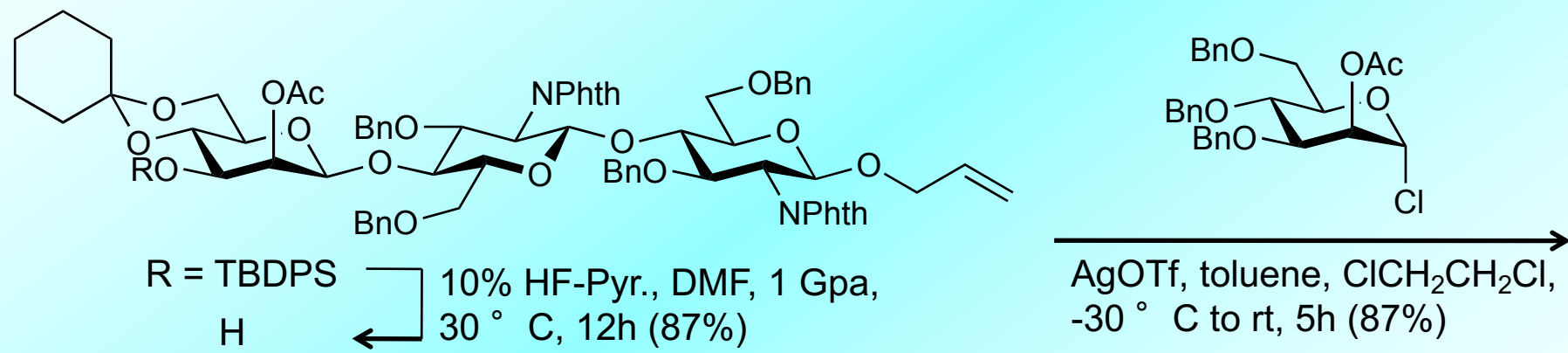
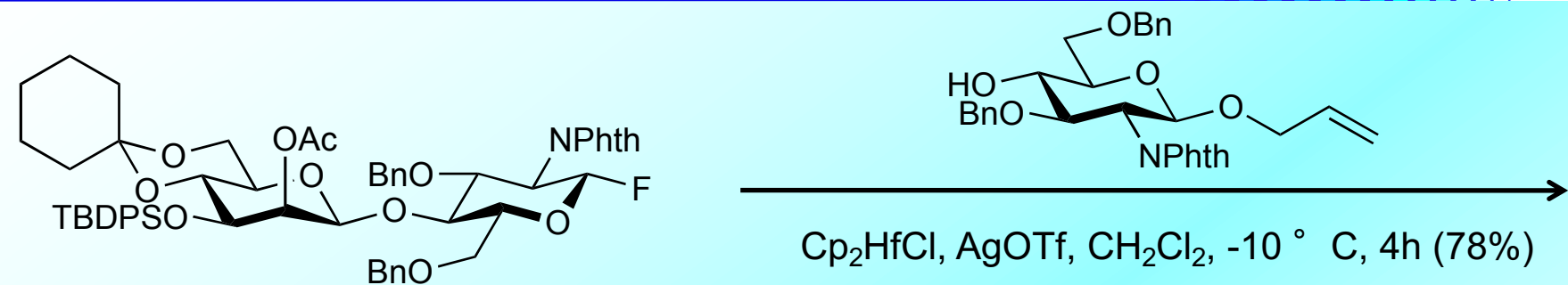
Chemical Formula: $C_{34}H_{58}N_2O_{26}$

Molecular Weight: 910.82

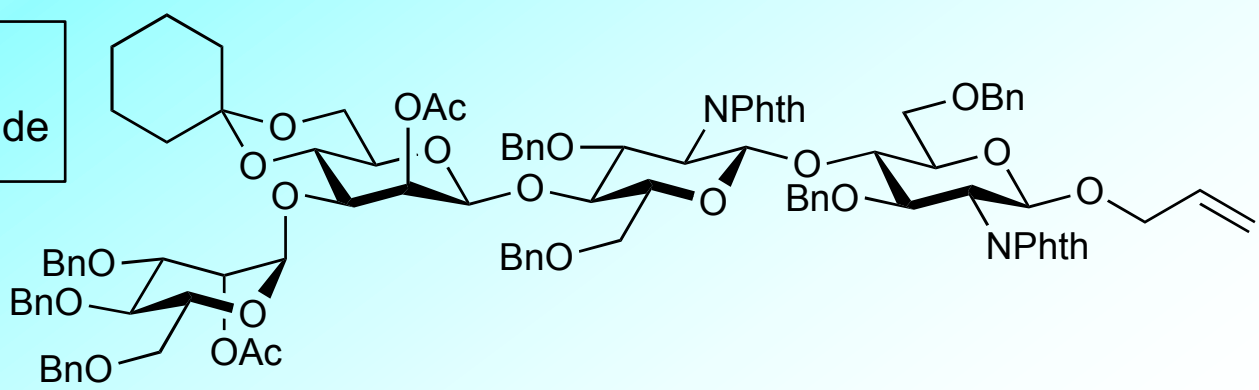
Synthesis of core pentasaccharide ~Part 1~



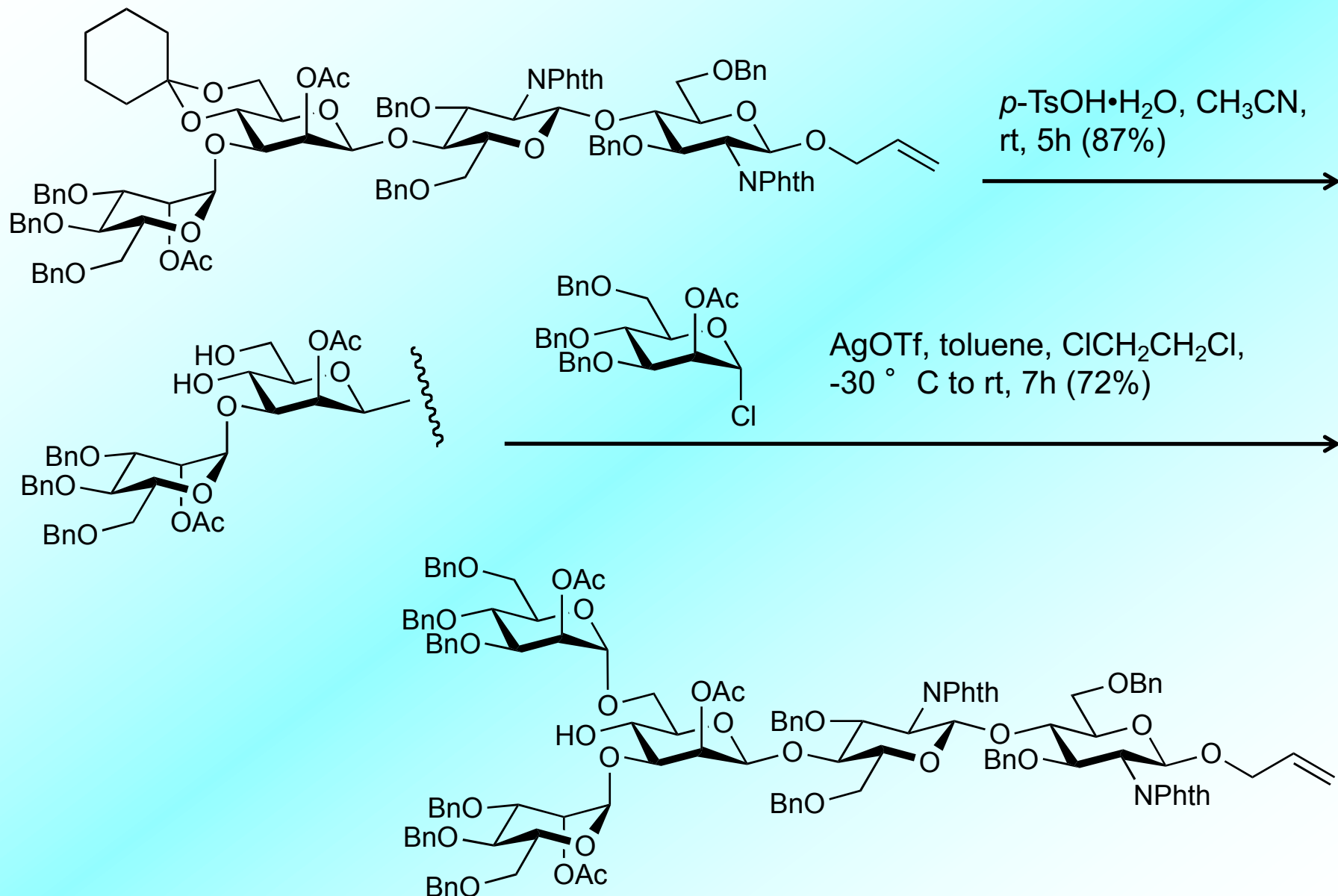
Synthesis of core pentasaccharide ~Part 2~



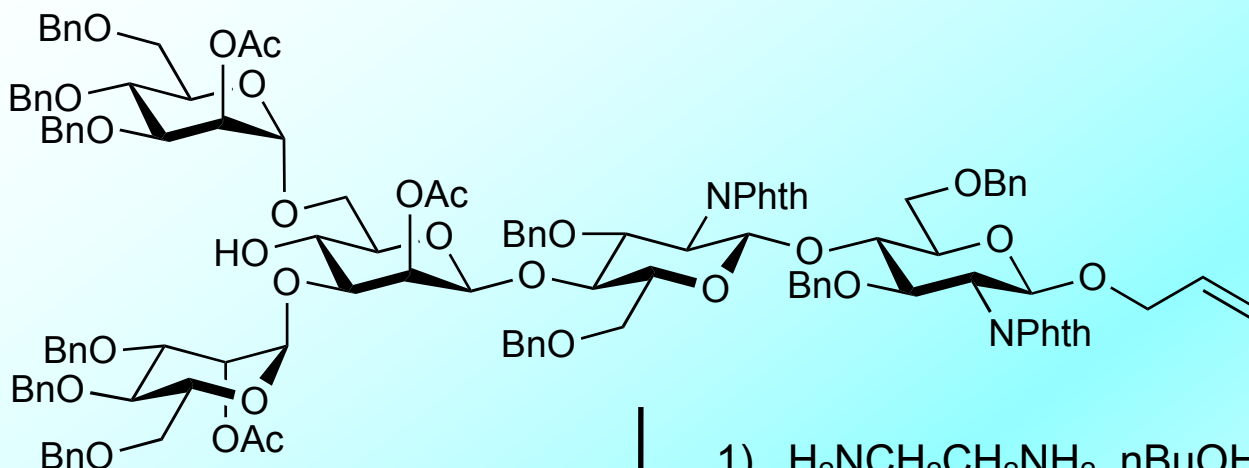
Cp = cyclopentadieny
DMF = N,N-dimethylformamide



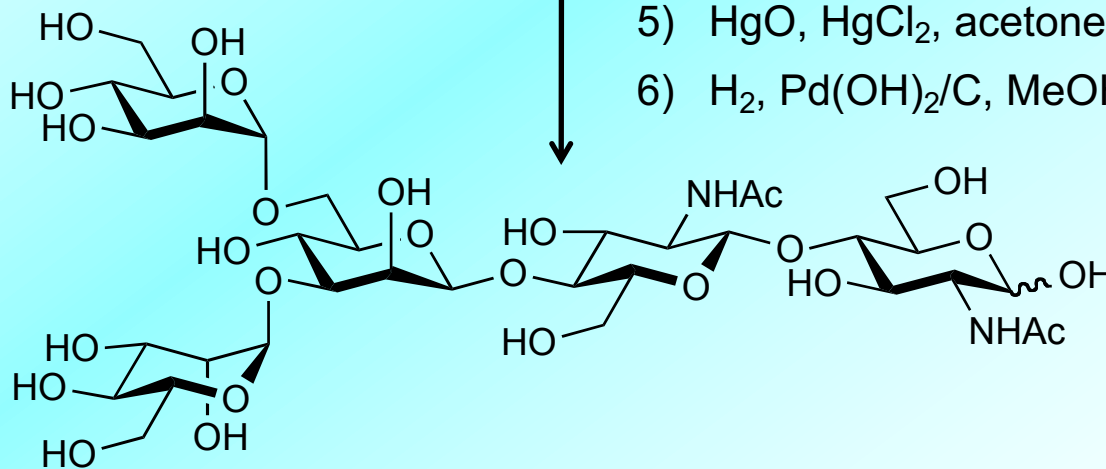
Synthesis of core pentasaccharide ~Part 3~



Synthesis of core pentasaccharide ~Part 4~



- 1) $\text{H}_2\text{NCH}_2\text{CH}_2\text{NH}_2$, nBuOH, 80 ° C, 18h
- 2) Ac_2O , pyr., DMAP, rt, 2.5h
- 3) 1M NaOMe, MeOH, rt, 3.5h (89%, 3 steps)
- 4) $[\text{Ir}(\text{COD})(\text{PMePh}_2)_2]\text{PF}_6$, H_2 , THF, rt, 2h
- 5) HgO, HgCl₂, acetone-H₂O (10:1), rt, 5h
- 6) H_2 , Pd(OH)₂/C, MeOH, rt, 2.5 h (85%, 3 steps)



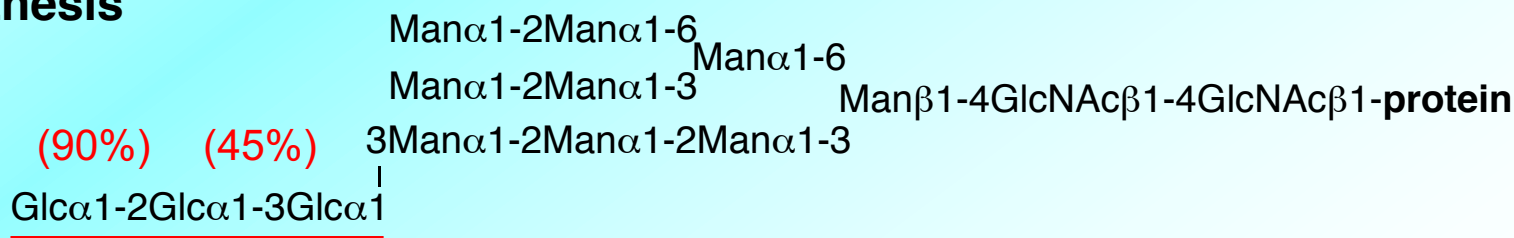


☒ Solvent Effect $\left\{ \begin{array}{l} \text{Yield: } 0 \sim 100\% \\ \alpha/\beta: 1:1 \sim 20:1 \end{array} \right.$

A. Ishiwata and Y. Ito *Tetrahedron Lett.* **2005**, 46, 3521.

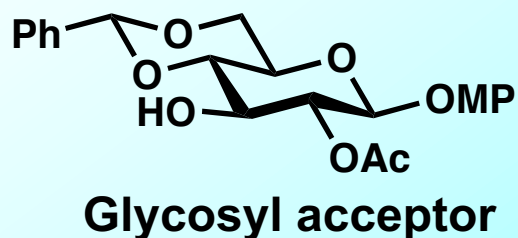
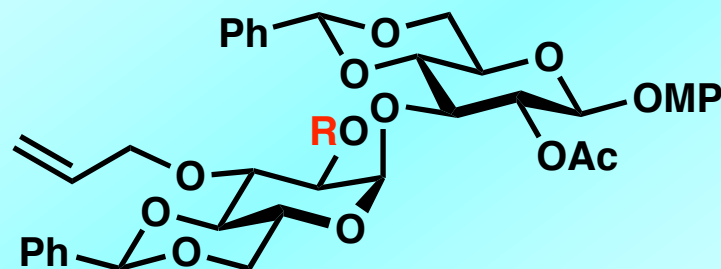
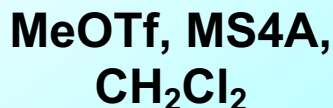
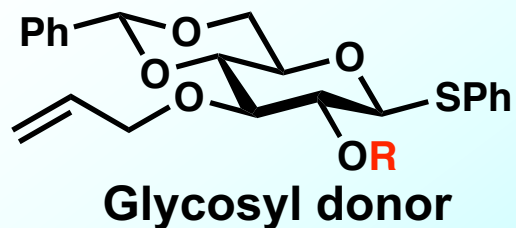
☒ High-mannose-glycan synthesis

I. Matsuo, K. Totani, A. Tatami and Y. Ito *Tetrahedron* **2006**, 62, 8262.

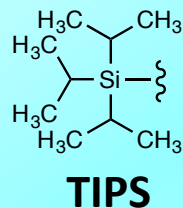
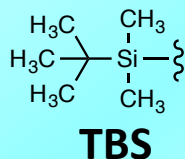
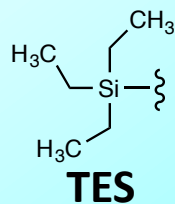
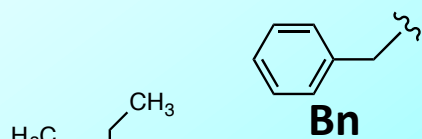


Development of 1,2-cis- α -glycoside formation reaction using stereoelectronic effect

K. Totani et al. RSC Advances 2015, 5, 75918.



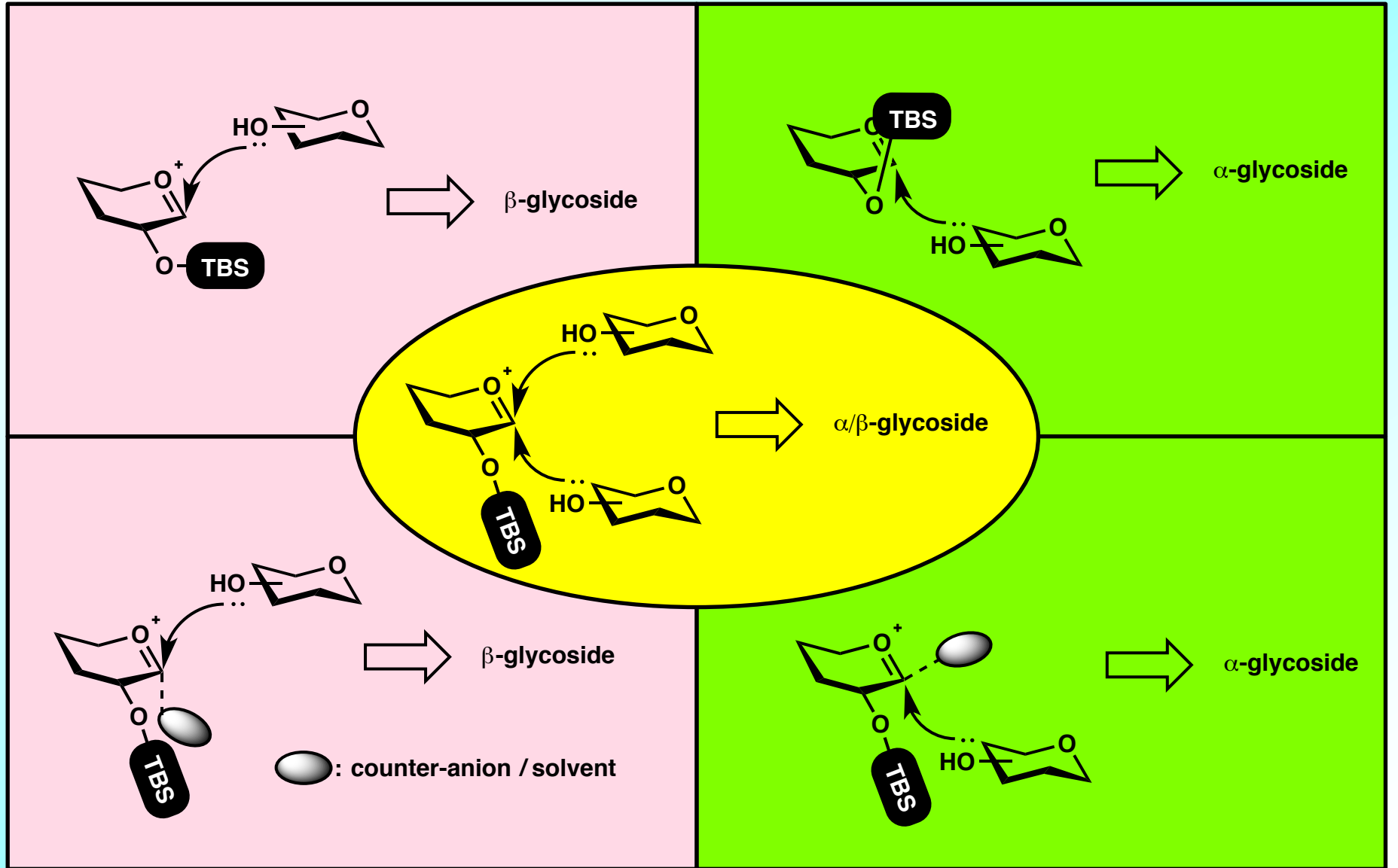
Entry	R	Temp. ($^{\circ}$ C)	Yield (%)	α / β
1	Bn	10	85	82 : 18
2	TES	0	-	-
3	TBS	0	96	>95 : 5
4	TIPS	0	50	>95 : 5



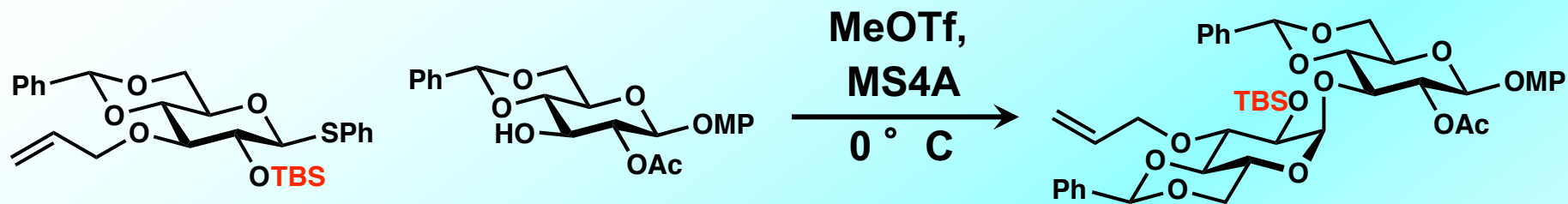
☒ **TBS group is effective for high yield and α -selectivity**

☒ **TBS effect : Steric effect or Electronic effect?**

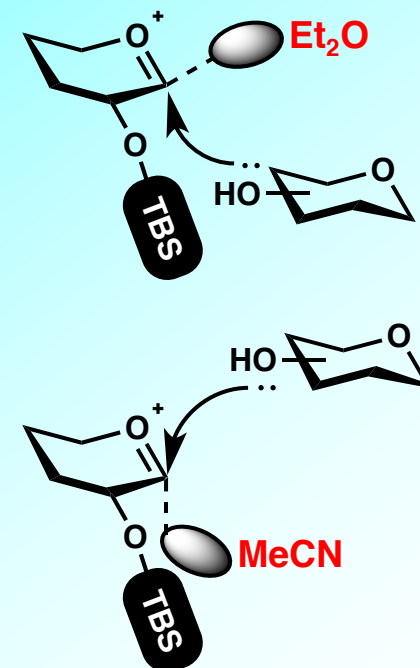
Possible stereocontrol modes



Estimation of stereocontrol mode based on the solvent effects

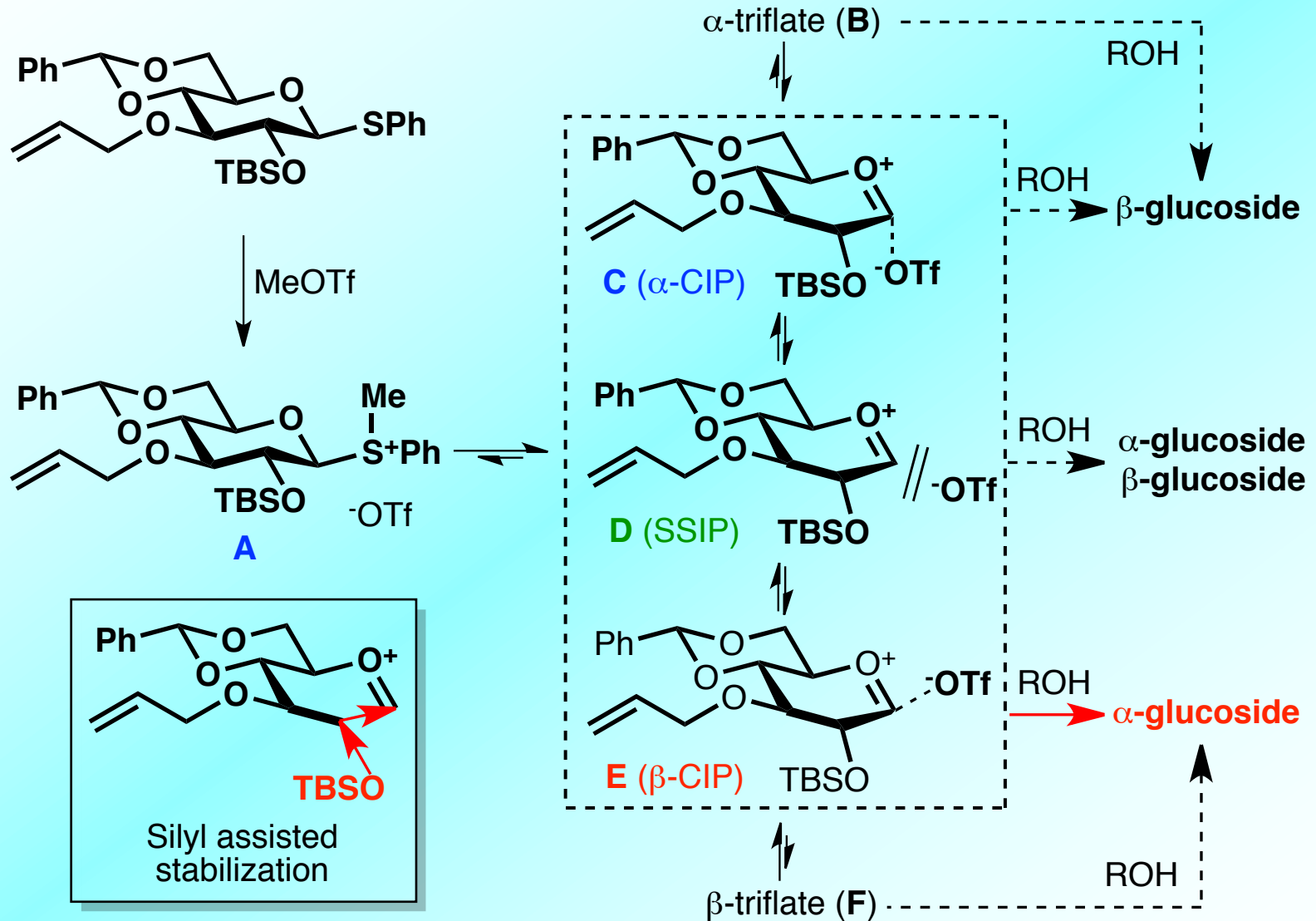


Entry	Solvent	Participation	Yield (%)	α / β
1	CH ₂ Cl ₂	None	96	>95 : 5
2	Toluene	None	41	>95 : 5
3	Et ₂ O	β	20	>95 : 5
4	MeCN	α	21	48 : 52



☒ **The steric bulkiness of TBS group does not affect the stereoselectivity**

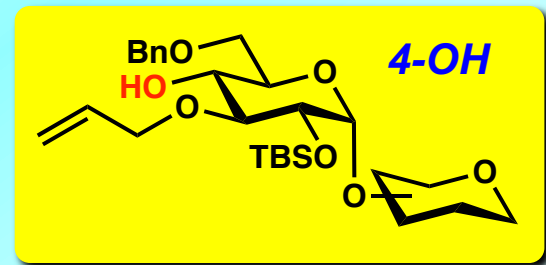
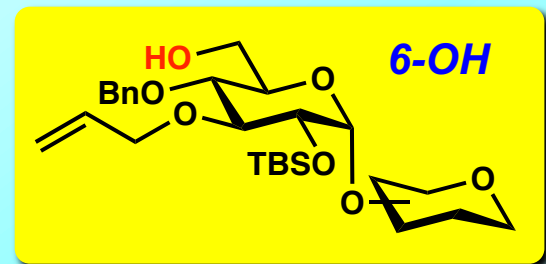
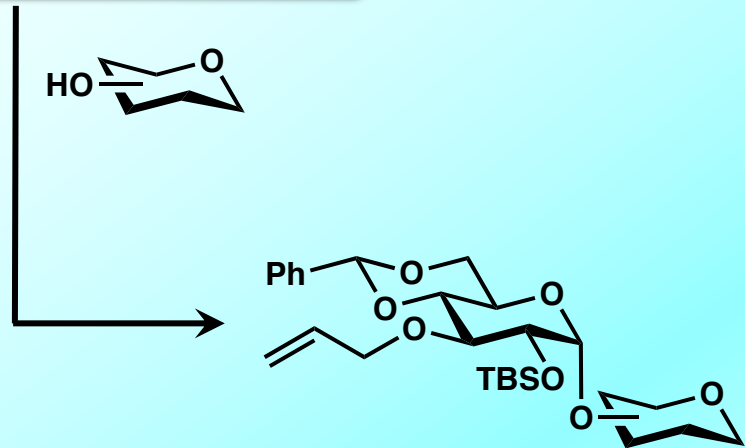
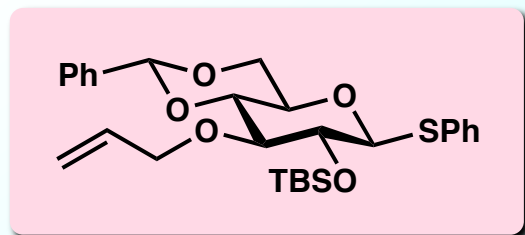
Stereocontrol mechanism of the α -glucosylation



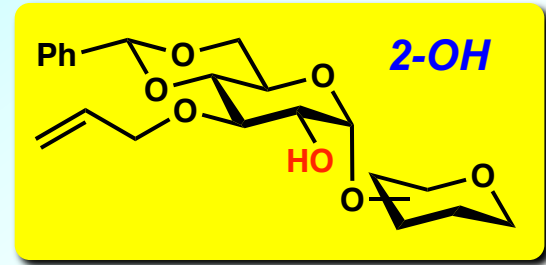
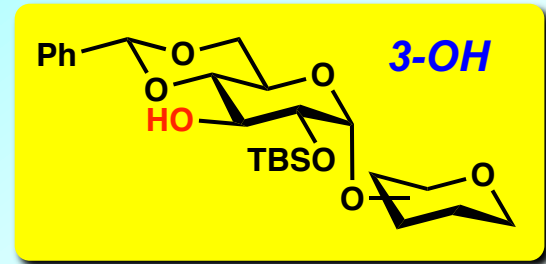
Versatility for glycosyl acceptors

Entry	Donor	Acceptor	Yield (%)	α / β
1			96	>95 : 5
2	↓		77	93 : 7
3			43	92 : 8
4		<i>Disarmed</i> 	82	>95 : 5
5		<i>Armed</i> 	96	>95 : 5

Advantages of the proposed glycosyl donor



Regioselective deprotection

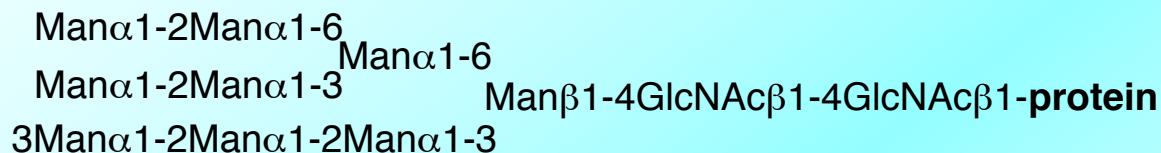
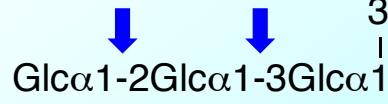


☒ **Further glycosylation is possible for all OH-groups**

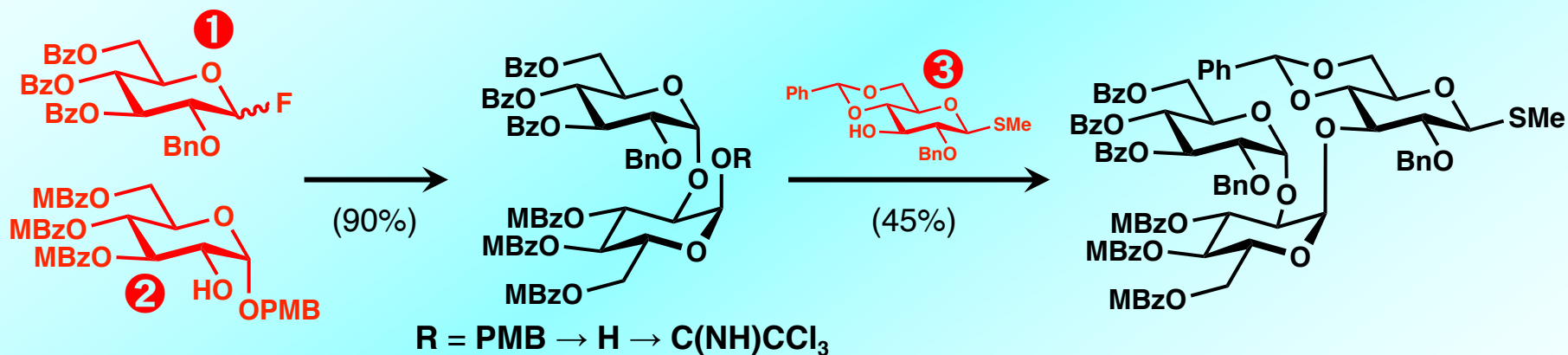
☒ **Effective for oligosaccharide synthesis**

Trisaccharide synthesis ~Existing approach~

1,2-*cis*- α -linkage



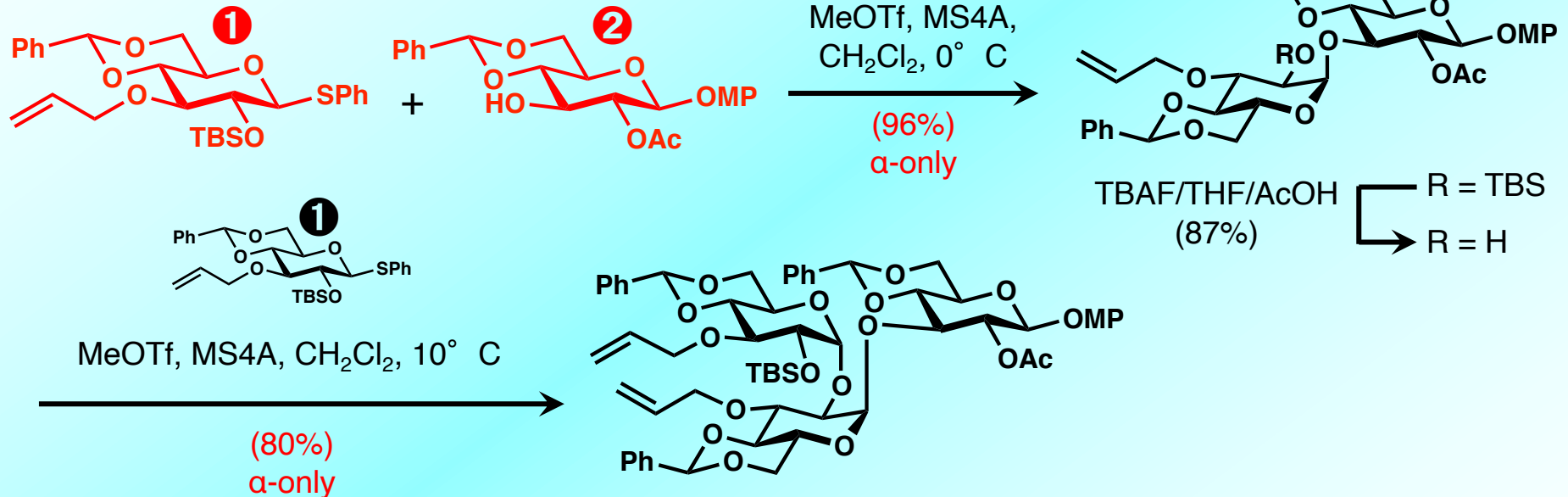
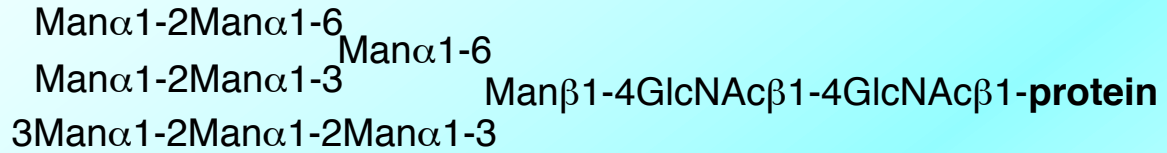
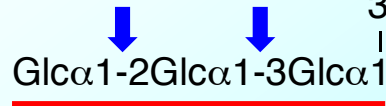
I. Matsuo, K. Totani, A. Tatami and Y. Ito *Tetrahedron* **2006**, 62, 8262.



- ☒ **Three types of monosaccharide units are required**
- ☒ **Four steps are required for conversion to target trisaccharide**
- ☒ **Low yield for the second glycosylation**

Trisaccharide synthesis ~Proposed approach~

1,2-cis- α -linkage

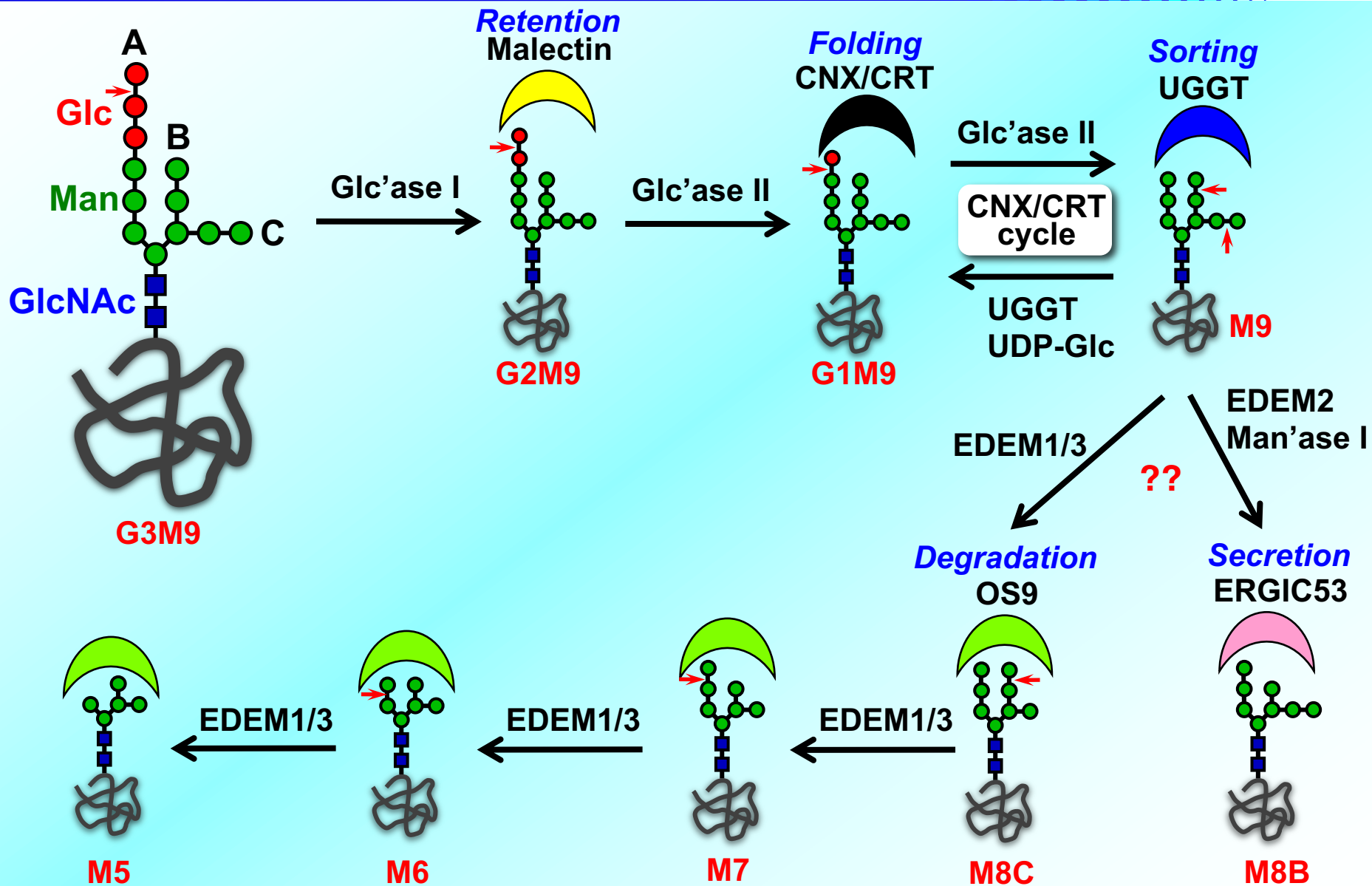


☒ **Monosaccharide units (3 \rightarrow 2)**

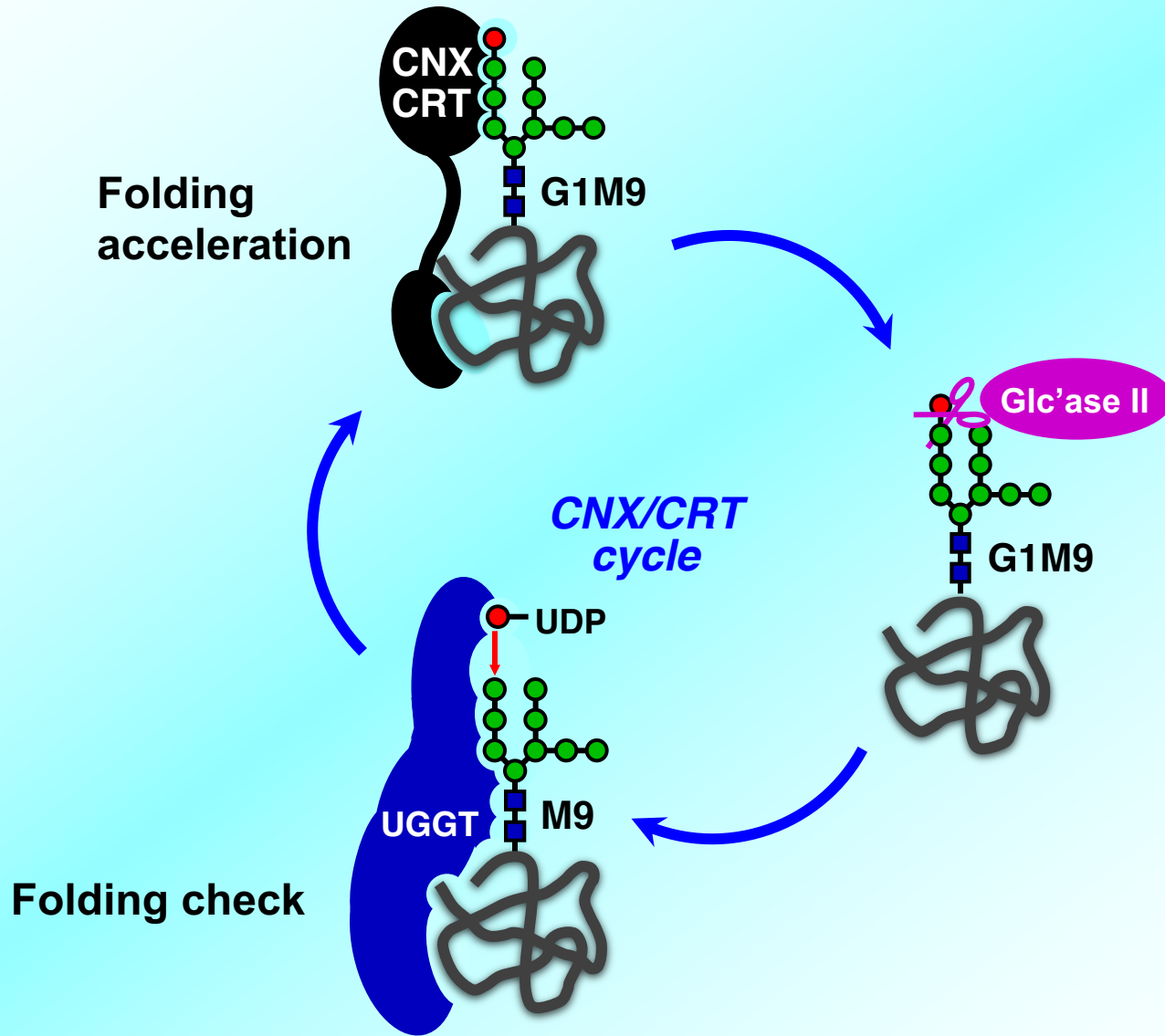
☒ **Synthetic steps (4 \rightarrow 3)**

☒ **Yield for the second glycosylation (Low \rightarrow High)**

Glycoprotein quality control

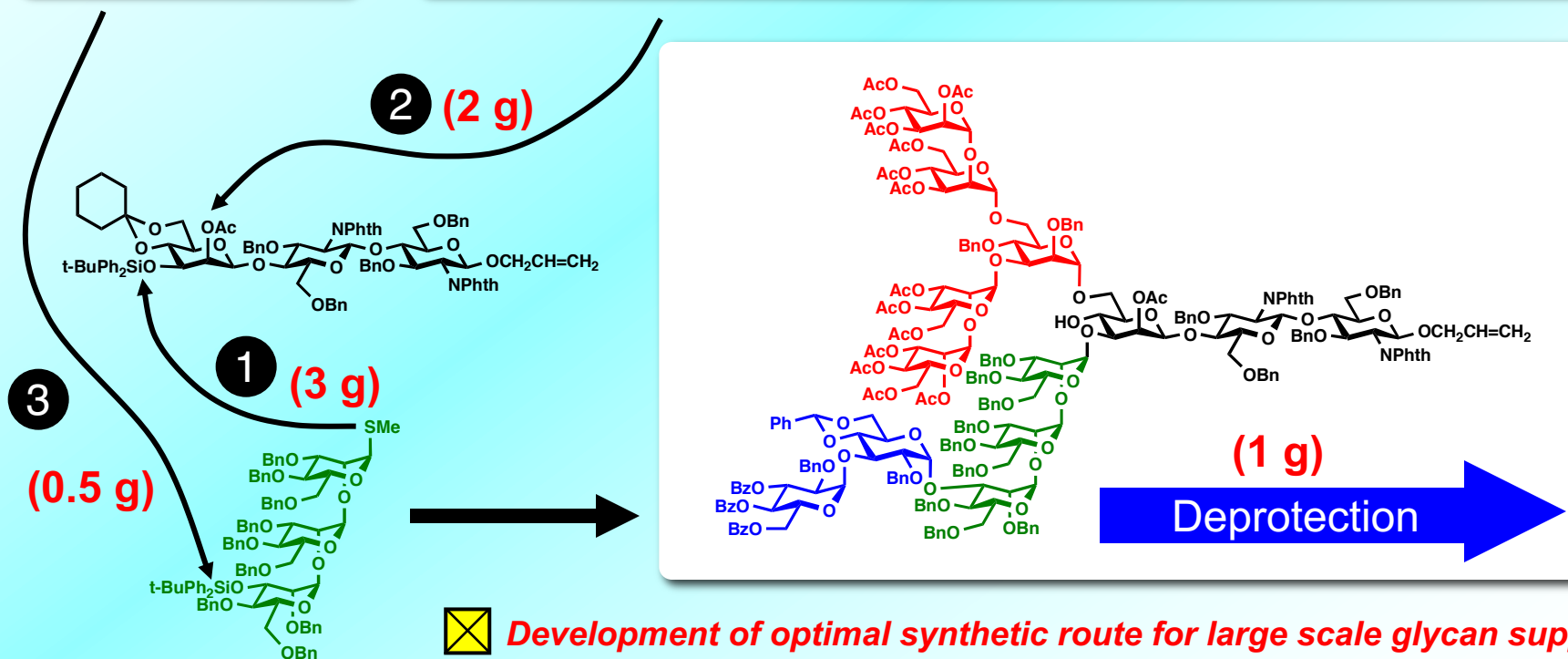
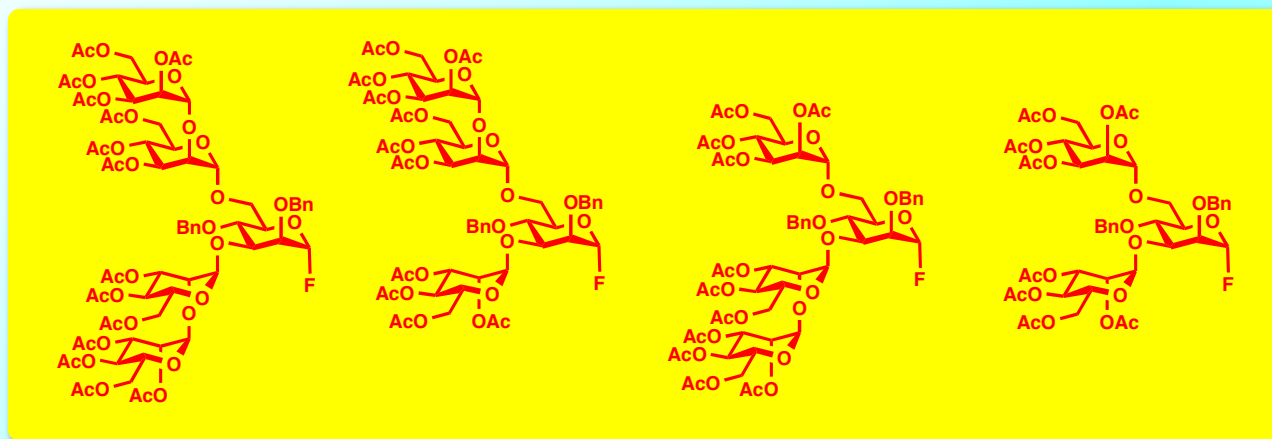
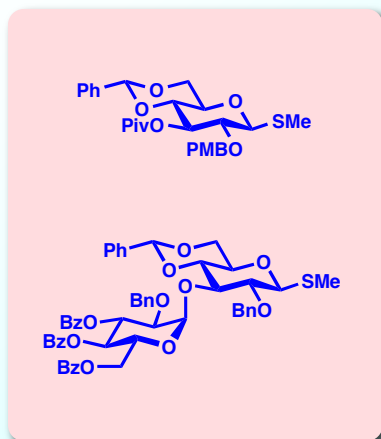


Central mechanism of glycoprotein quality control ~CNX/CRT cycle~



Synthesis of substrate glycans

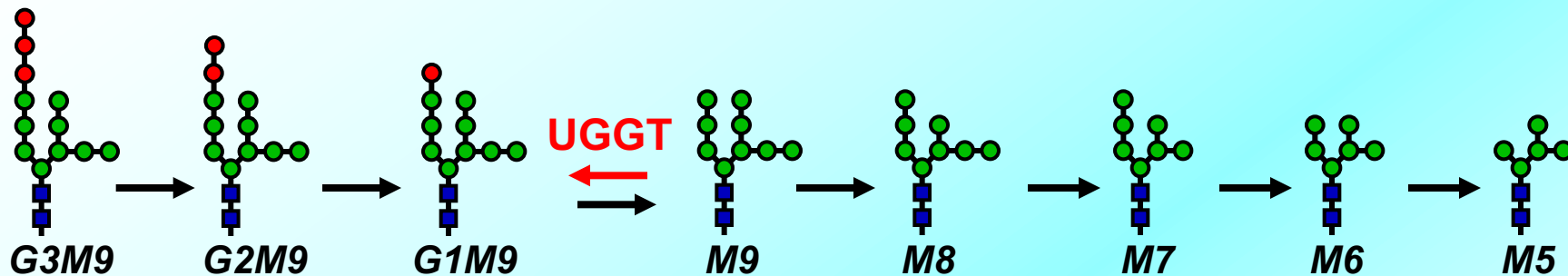
I. Matsuo et al. *Tetrahedron* 2006, 62, 8262.



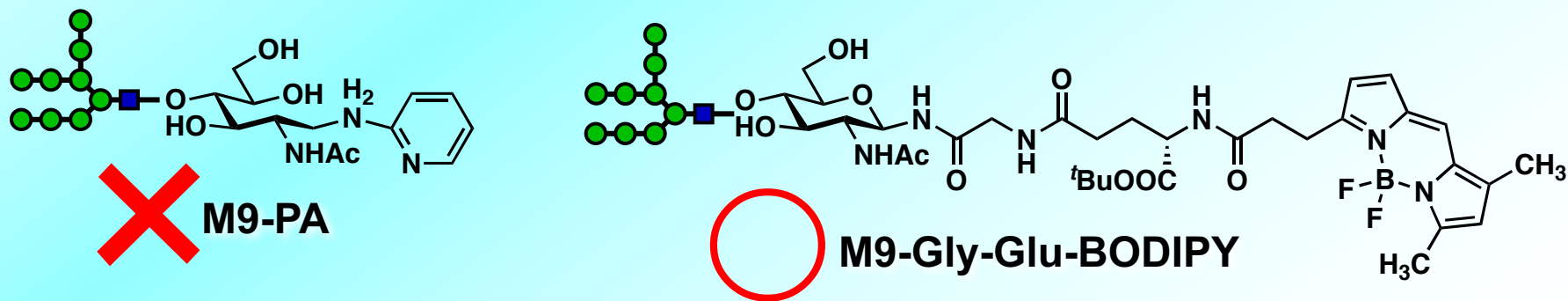
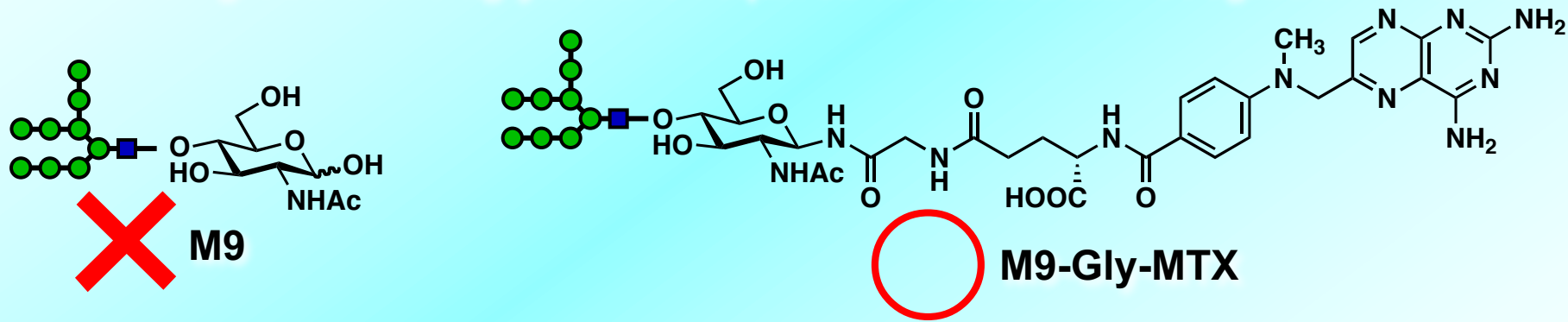
What is the best substrate for glycan processing analysis?

K. Totani et al. *Angew. Chem. Int. Ed.* 2005, 44, 7950.

K. Totani et al. *Biochemistry* 2009, 48, 2933.



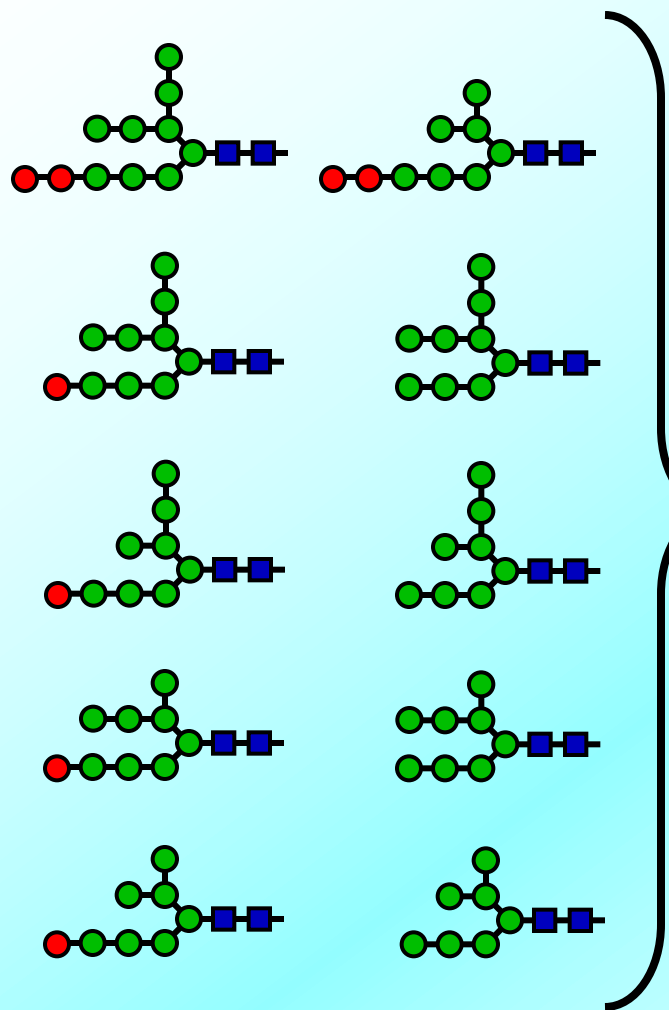
☒ **Molten globule-like aglycone is required for substrate recognition of UGGT**



Examples of synthetic substrates

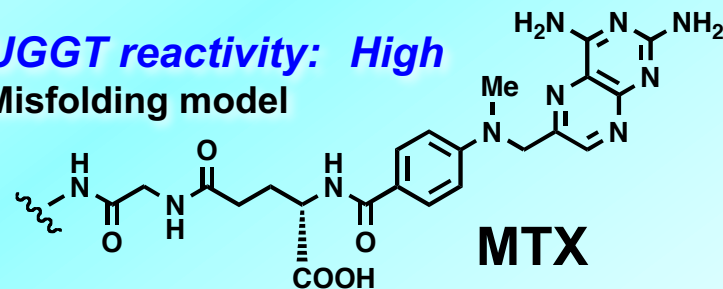
K. Totani et al. *Angew. Chem. Int. Ed.* 2005, 44, 7950.

K. Totani et al. *Bioorg. Med. Chem.* 2006, 14, 5220.

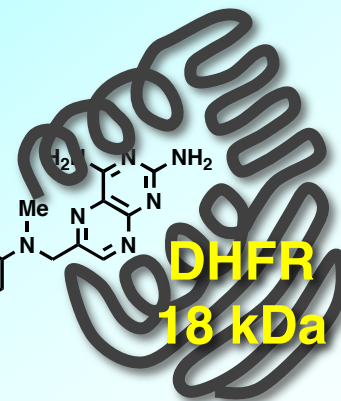
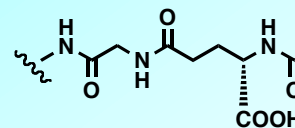


-OH **UGGT reactivity: none**

UGGT reactivity: High
Misfolding model



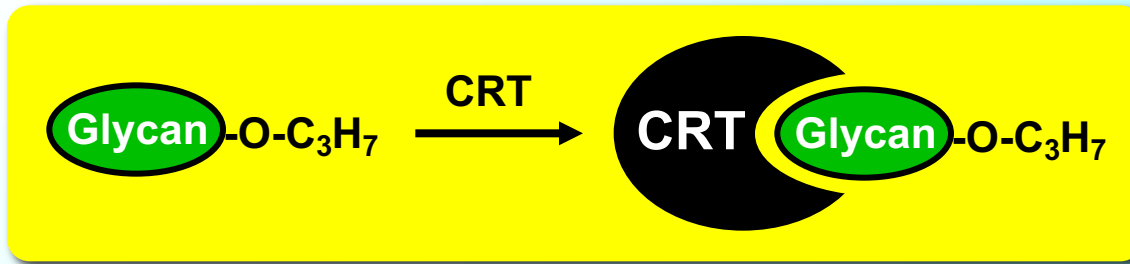
UGGT reactivity: Low
Proper folding model



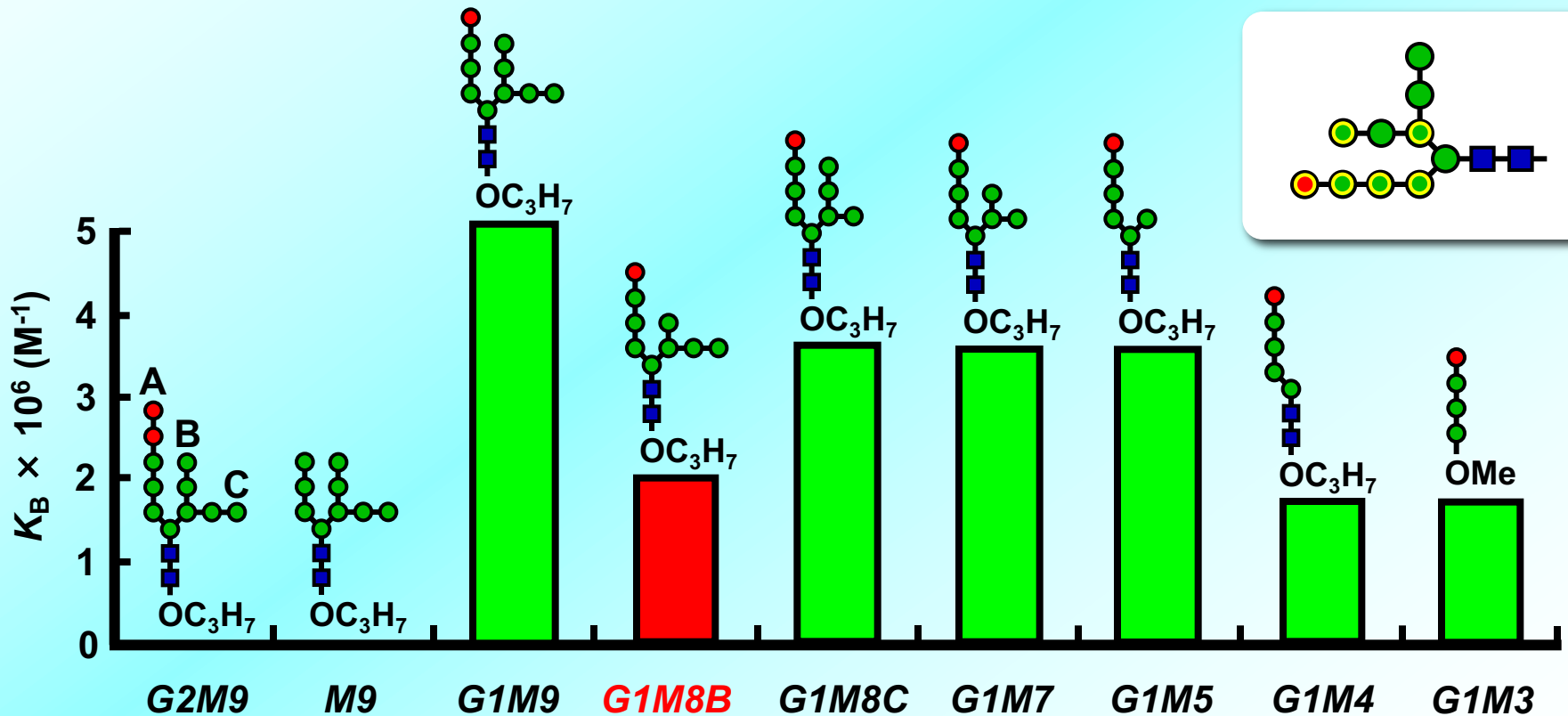
10 glycans × 3 functions = 30 substrate

Glycan specificity of CRT

Y. Ito et al. *Curr. Opin. Struct. Biol.* 2005, 15, 481.



- ☒ *Specific recognition of G1 glycans*
- ☒ *Low affinity for G1M8B*
- ☒ *Branch Man for B-, C-arm is essential*

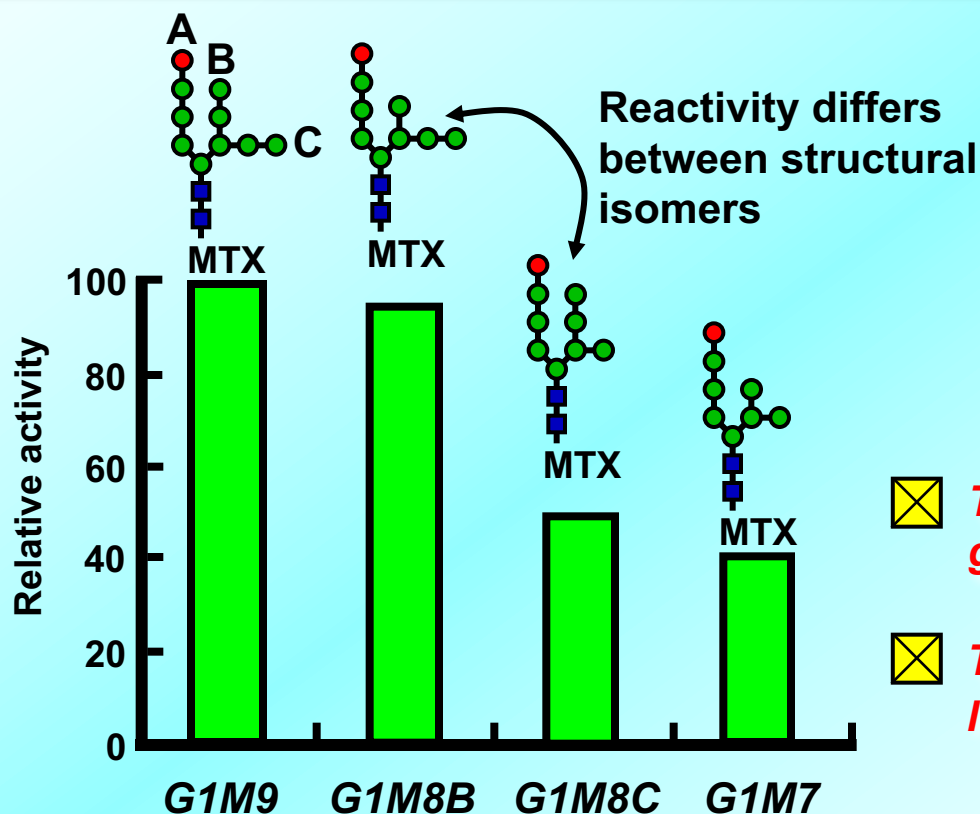


Glycan specificity of Glc'ase II

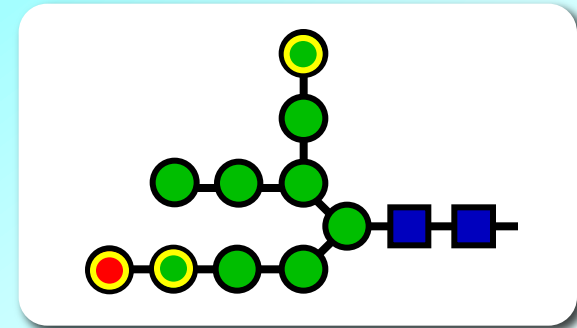
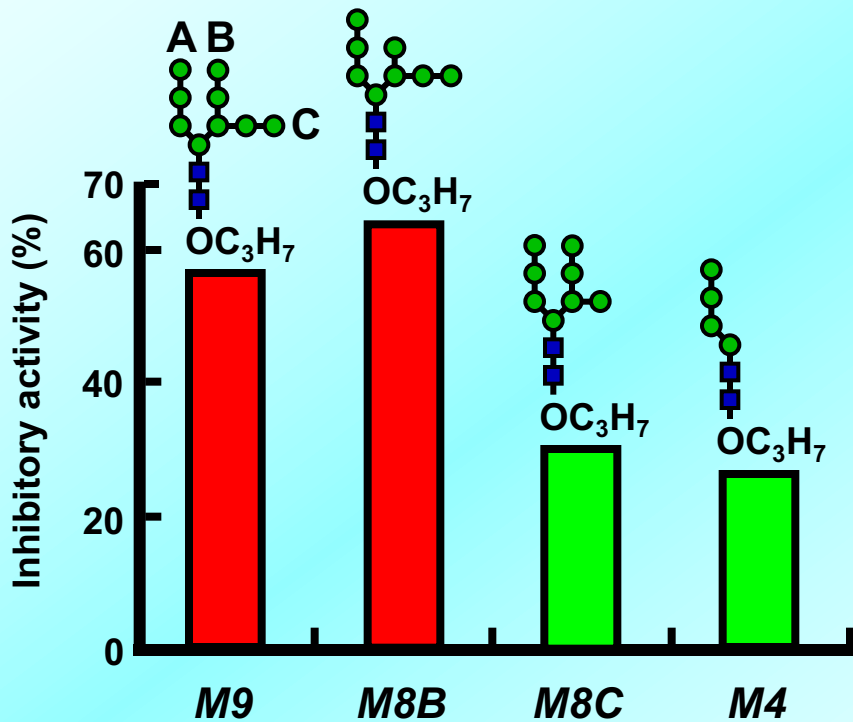
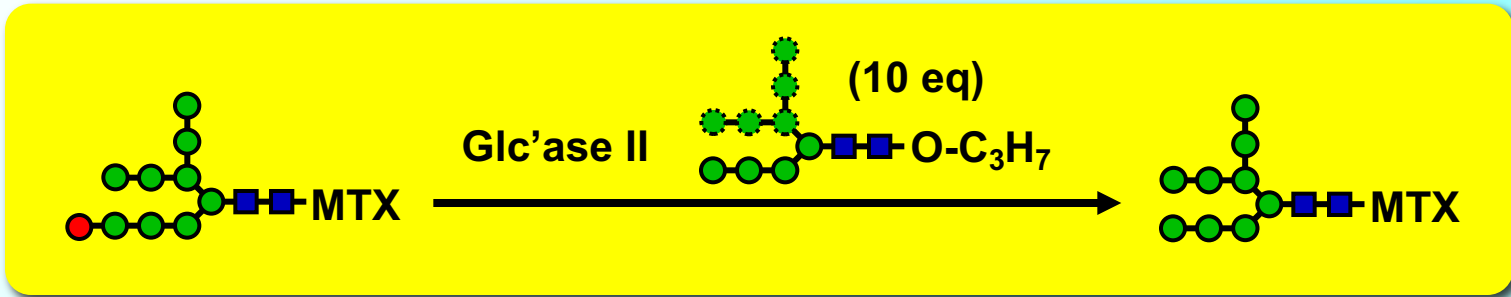
K. Totani et al. J. Biol. Chem. 2006, 281, 31502



Substrate	K_m (μM)	V_{max} ($\mu\text{mol/h/mg}$)
G1M9-MTX	78	7.87
G1M8B-MTX	56	6.78
G1M8C-MTX	93	5.26
G1M7-MTX	102	5.25



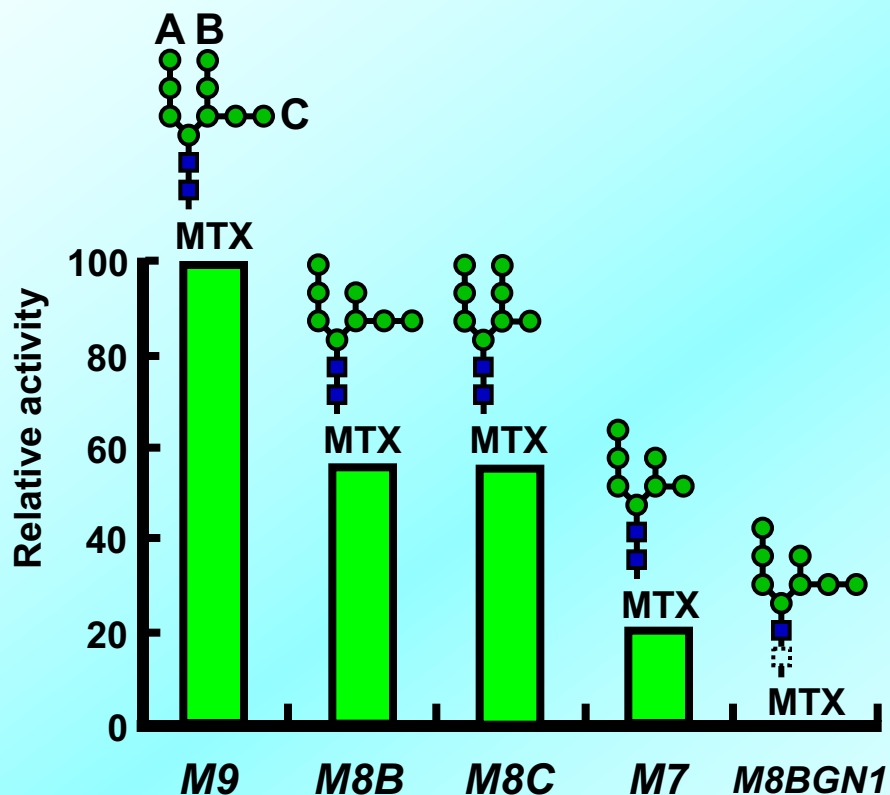
- ⊠ Terminal Man trimming at the B-arm gives little influence on the activity
- ⊠ Terminal Man trimming at the C-arm leads to reduced reaction efficiency



 **The terminal Man at the C-arm is important for the recognition**

Glycan specificity of UGGT

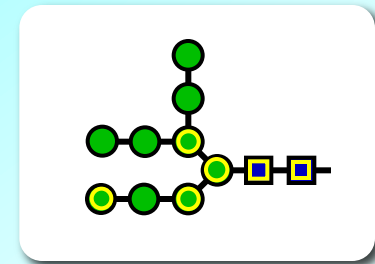
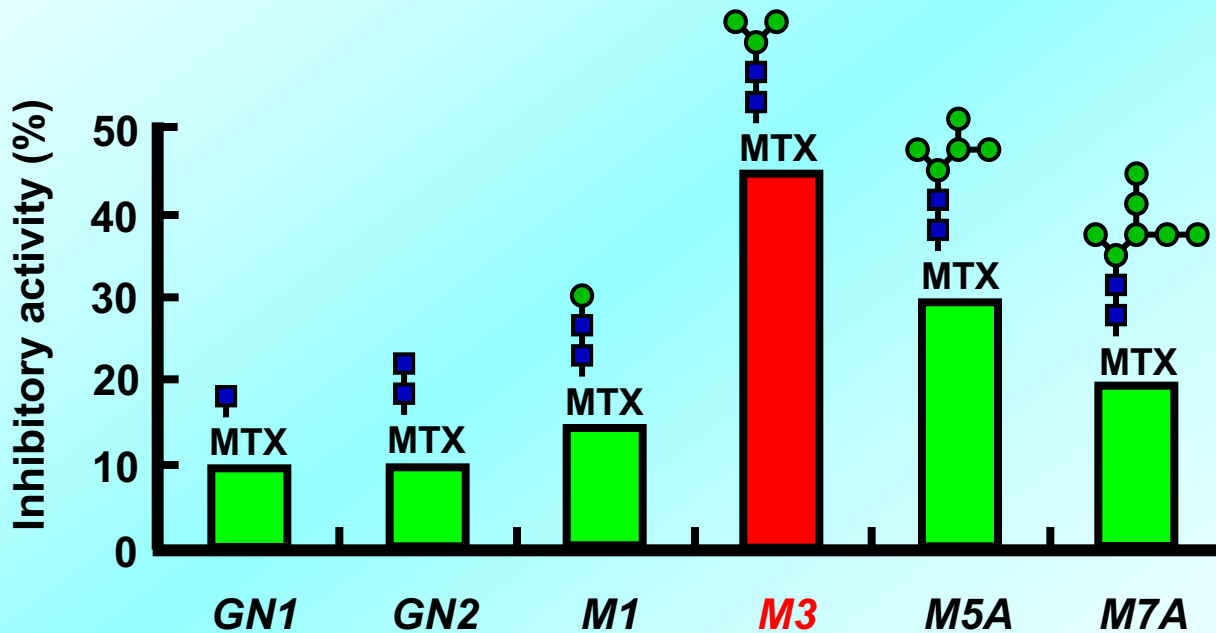
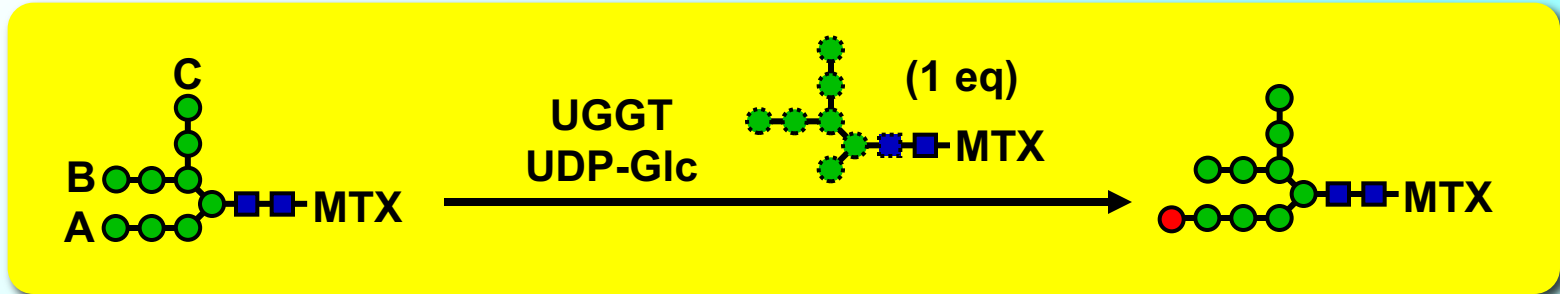
K. Totani et al. *Angew. Chem. Int. Ed.* 2005, 44, 7950.



Substrate	K_m (μM)	V_{max} (nmol/h/mg)
M9-MTX	207	32.3
M8B-MTX	197	15.4
M8C-MTX	448	23.8
M7-MTX	46	4.8

⊠ Terminal Man trimmings at the B- and C-arm lead to reduced reaction efficiency

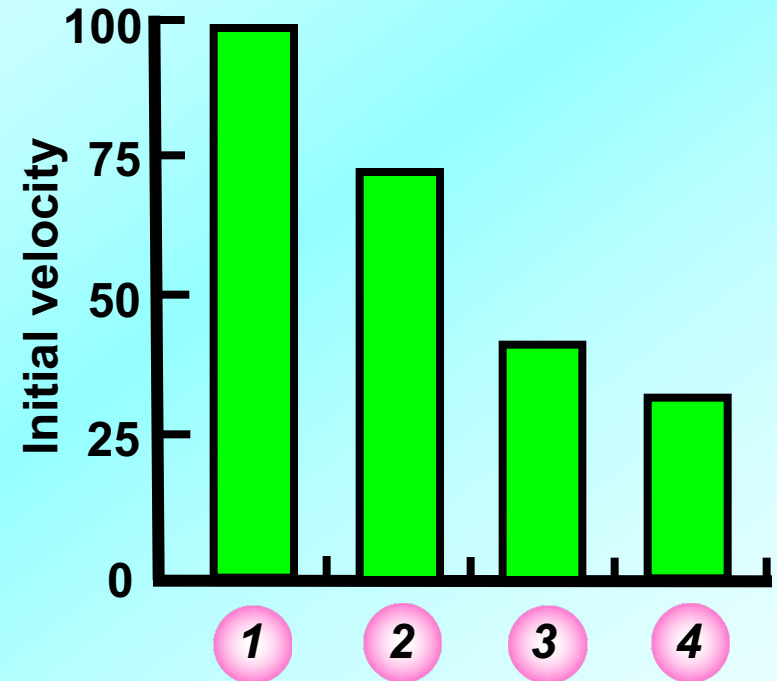
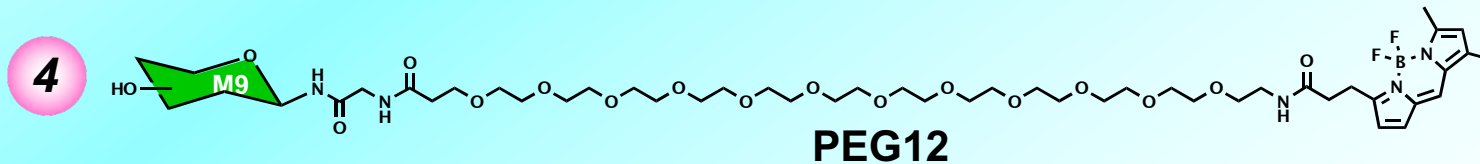
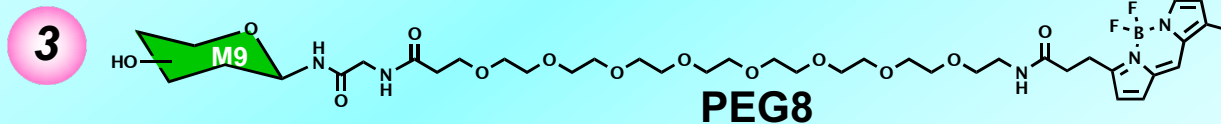
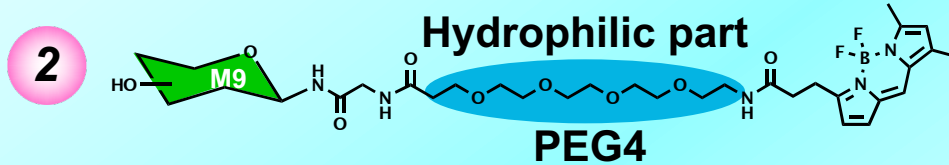
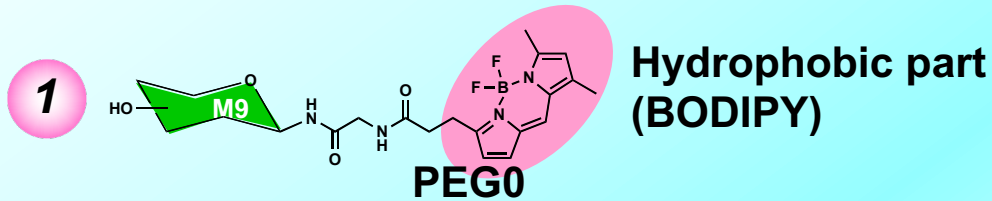
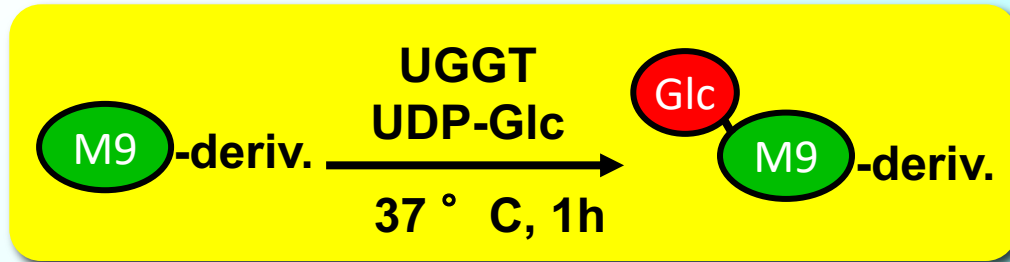
⊠ GlcNAc2 is required for Glc-transfer




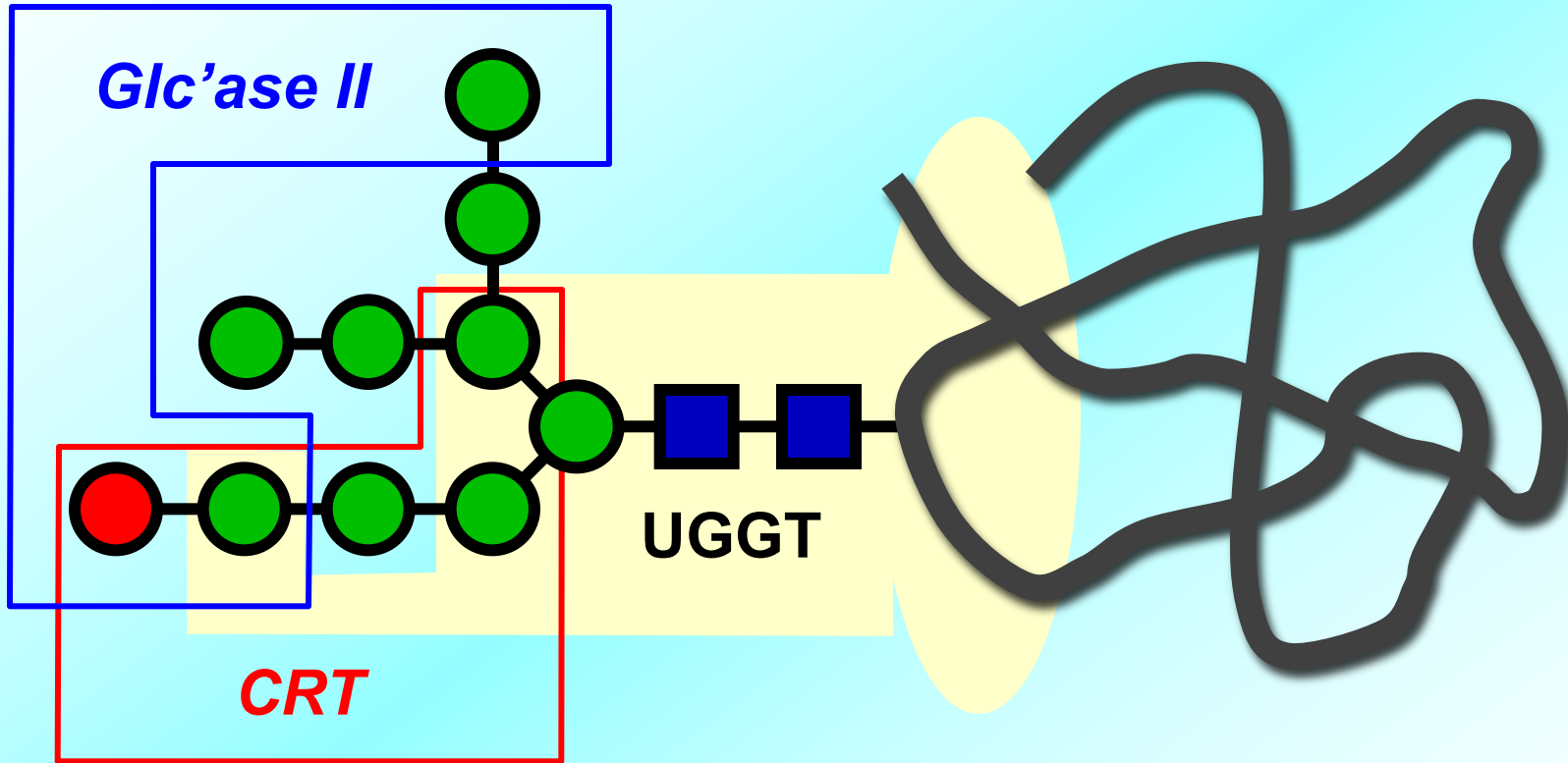
Man3GlcNAc2 structure is mainly recognized

Aglycone specificity of UGGT

K. Totani et al. *Biochemistry* 2009, 48, 2933.



 **Substrate recognition ability decreases as increasing hydrophilicity near the reducing end**



☒ *Quality control that skilfully uses the recognition part*